

De-excitation spectroscopy II: Photon-in Photon-out spectroscopy

Excitation Source: Tunable SR (UV –X rays)

Photon-out phenomena:

- *Scattering* (elastic, inelastic, resonant)
- *Fluorescence* (core-hole decay)
 - X-ray fluorescence (hard x-rays)
 - X-ray emission (soft x-rays)
 - Luminescence (UV-visible)
 - Auger (pseudo-photon)

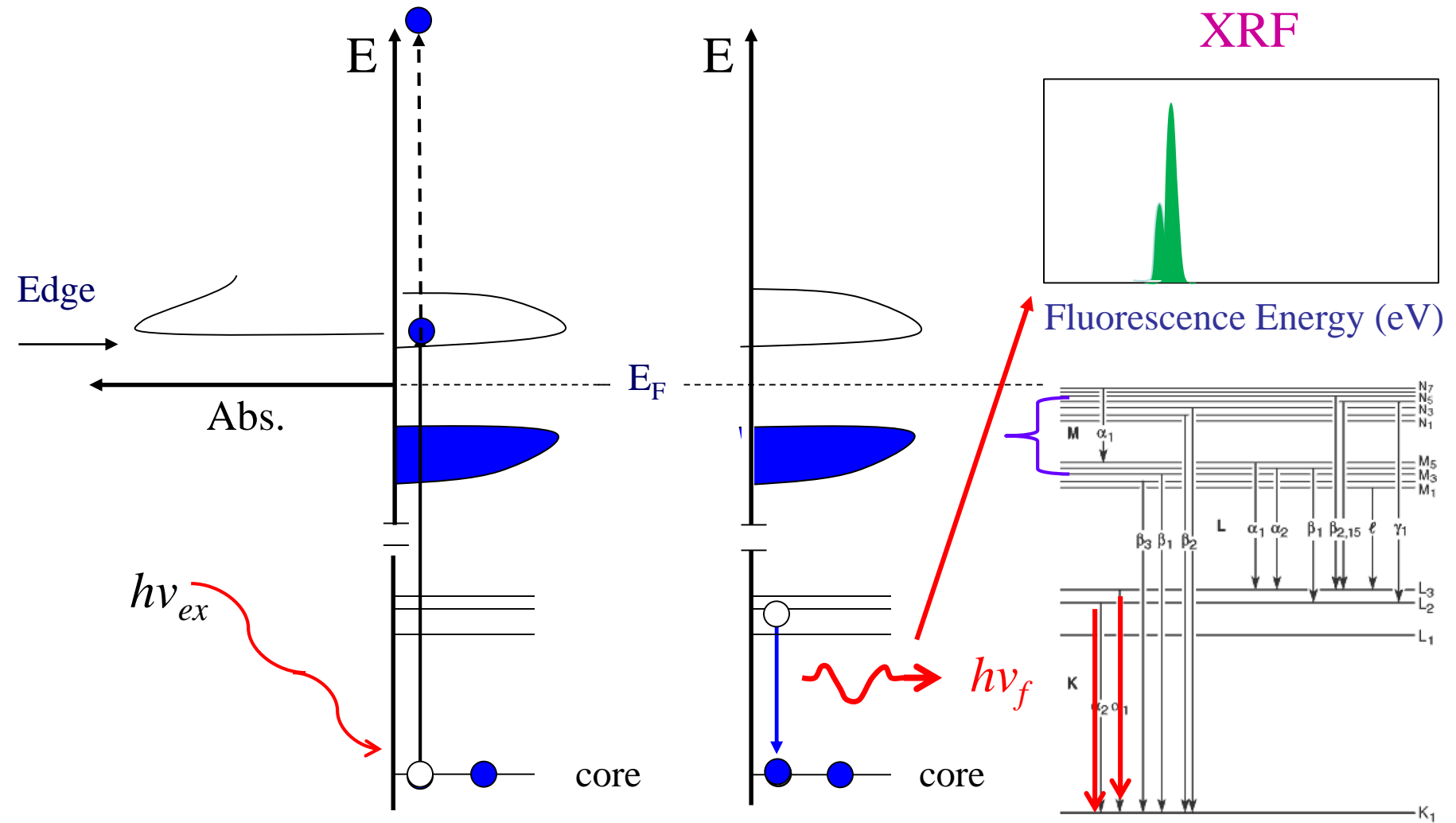
Photon-in Photon-out Spectroscopy

X-ray fluorescence (XRF)

X-ray emission (XES)

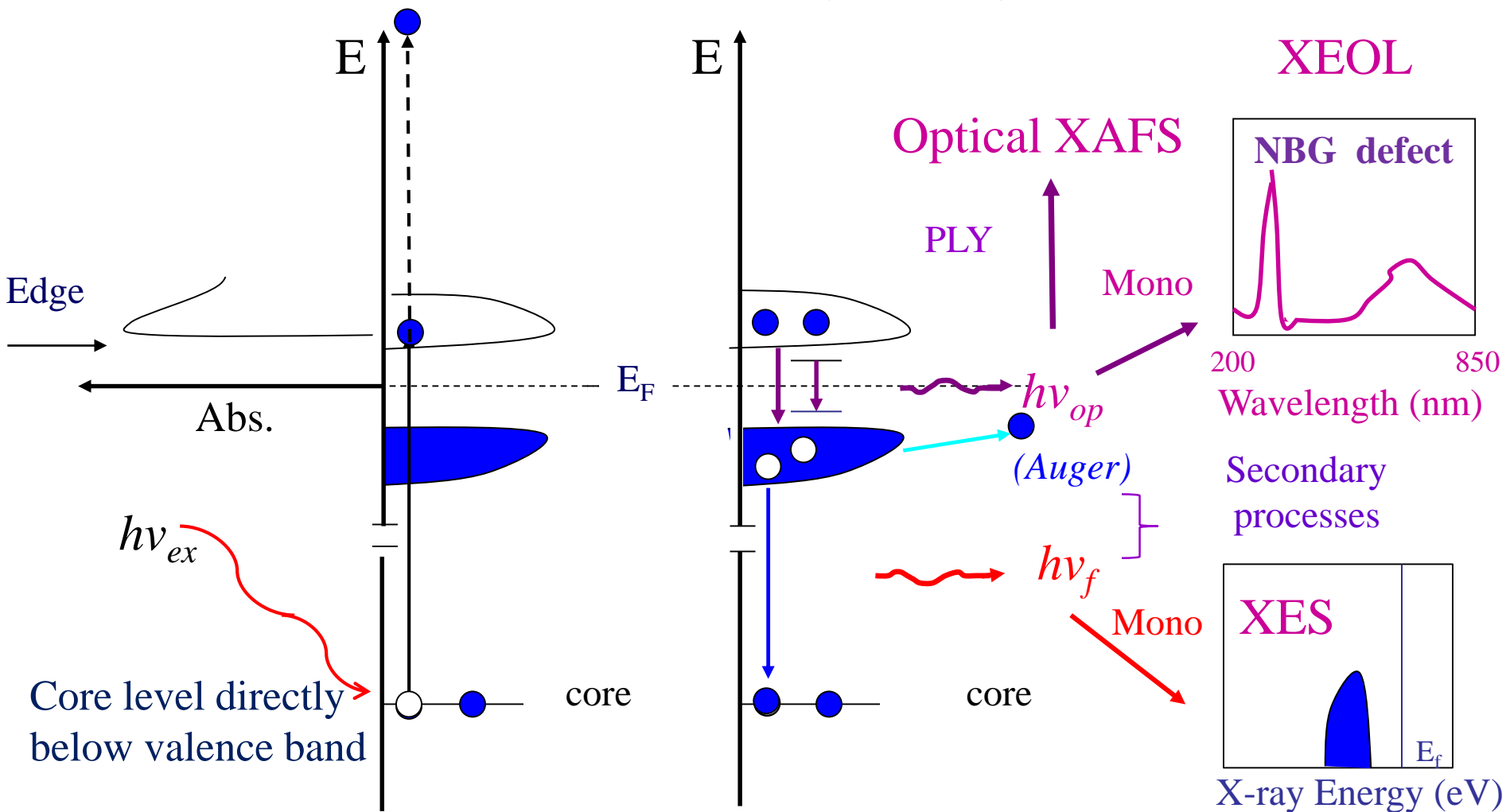
X-ray excited optical luminescence (XEOL)

X-ray fluorescence (XRF)



X-ray fluorescence: shallower **core** electron (e.g. L shell) to deeper **core** hole (K) transition

X-ray emission (XES) and X-ray excited optical luminescence (XEOL)



XES: valence electron to shallow core

XEOL: CB to VB & defects

X-ray Fluorescence measurements

Nondispersive (no energy resolution)

- Scintillation counter
- Ion chamber (Lytle detector) (with filters)

Moderate energy resolution

Solid state detectors (Ge, Si), order of 10 to 10^2 eV

High energy resolution

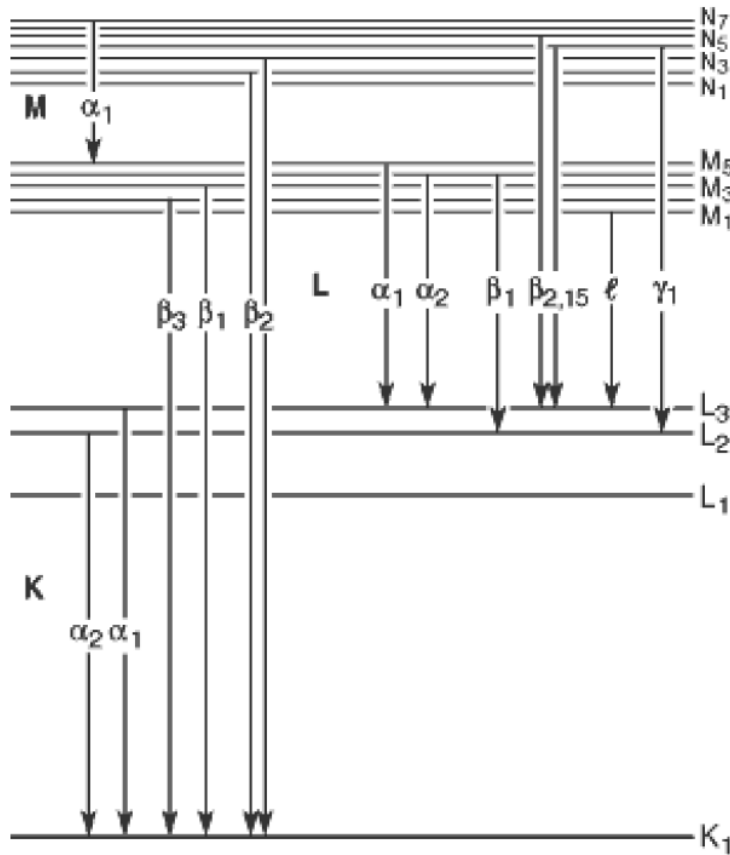
WDX detector (crystal monochromator) $E/\Delta E \sim 800$

Very high energy resolution

MiniXS (Jerry Seidler, U. of Washington)

$E/\Delta E \sim 4000$

X-ray Fluorescence properties of elements



Normal Fluorescence: Core-Core
XES: L or M = valence band
Resonant: Core-CB excitation

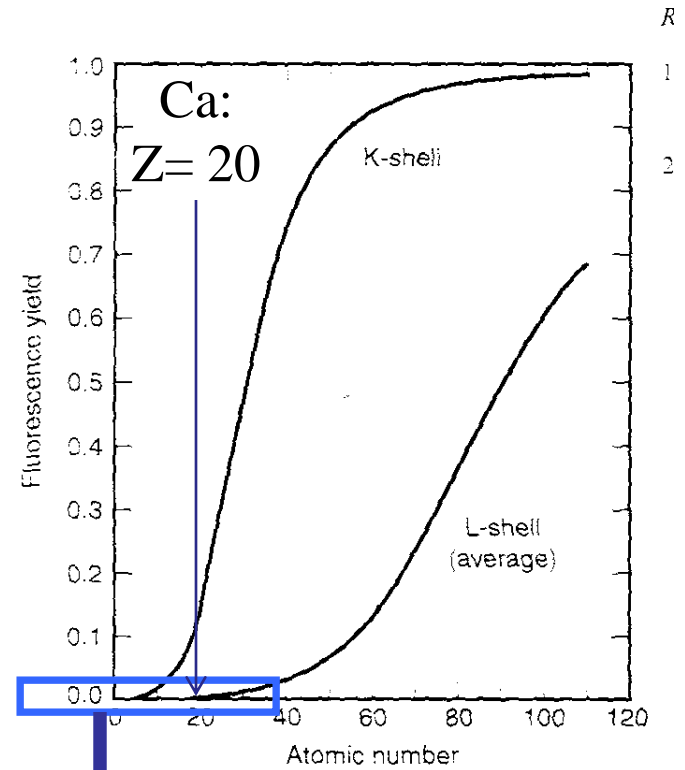


Fig. 1-2. Fluorescence yields for K and L shells for $5 \leq Z \leq 110$. The plotted curve for the L shell represents an average of L₁, L₂, and L₃ effective yields.

Auger yield = 1 – FLY
FLY for low z elements
(C, N, O etc.) is << 1 %

X-ray Fluorescence:

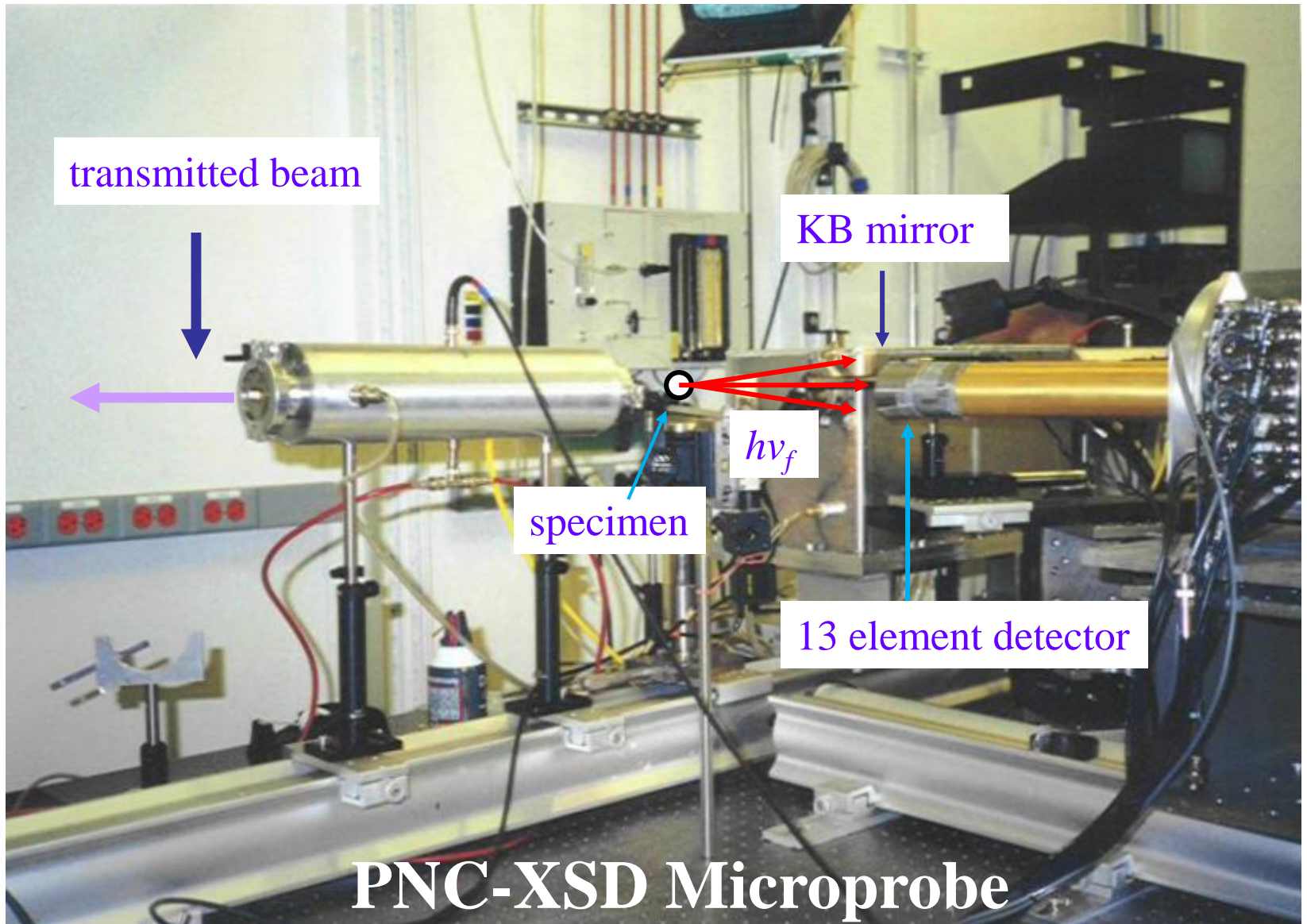
X-ray photons in, x-ray photons out.

Results from the decay of a deep core hole (e.g. K, or L)

Monitor the absorption spectrum using fluorescence yield (**FLY**) → **element specificity**

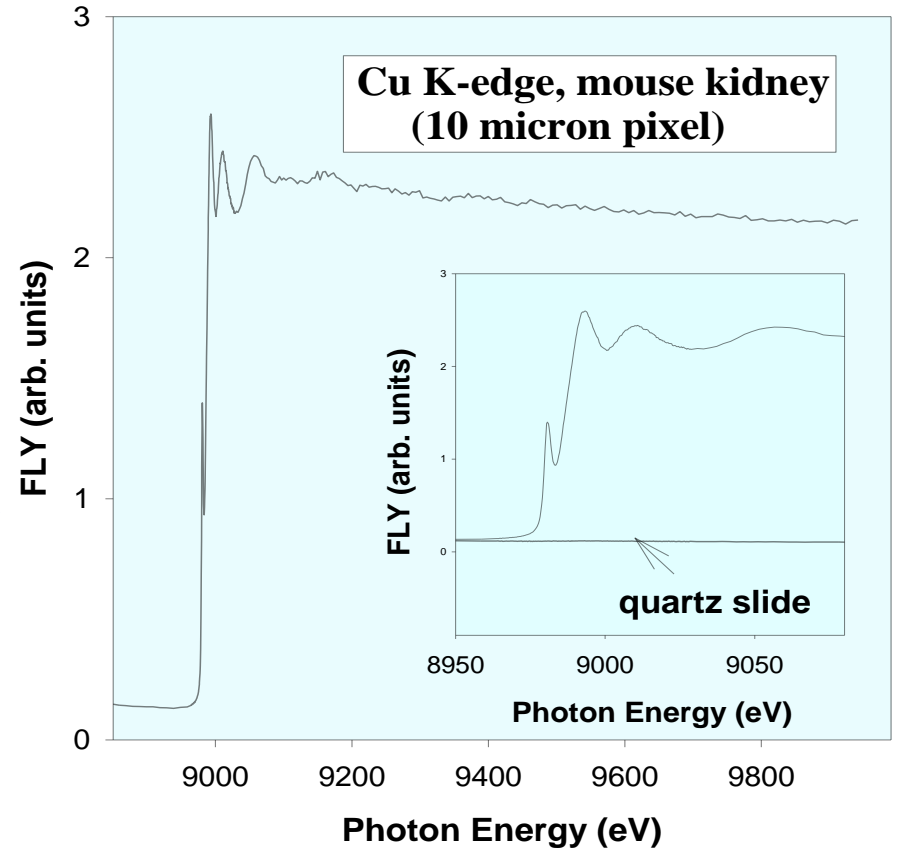
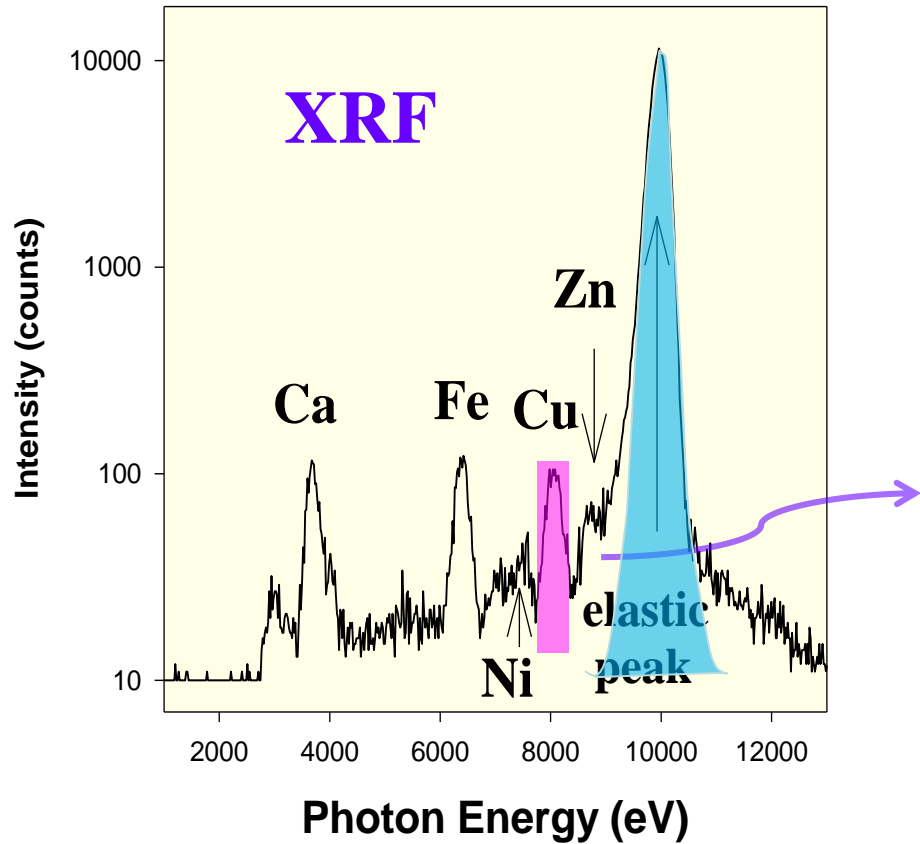
Detection Schemes

- High sensitivity, non-dispersive
Channel plate, Ion chamber
(solar slits, filters, e.g. Lytle)
- High sensitivity, moderate energy resolution
Solid state detector (e.g. Canberra 13 element
at PNC-CAT, Si drift detector at CLS)
Gas proportional counters (e.g. Fisher/Ohta)
- High sensitivity, more moderate resolution
Multi-layer Array Analyzer Detector (MAAD)
Log spiral detector (asymmetric Laue bent)

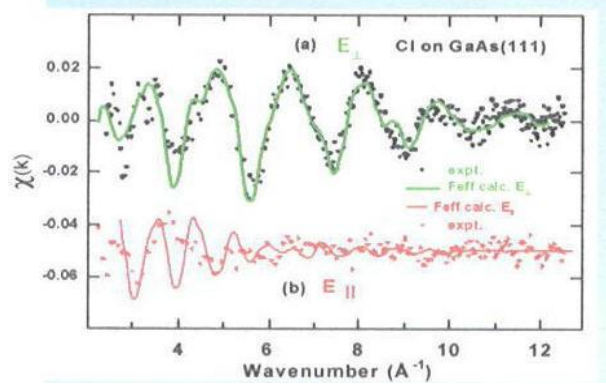
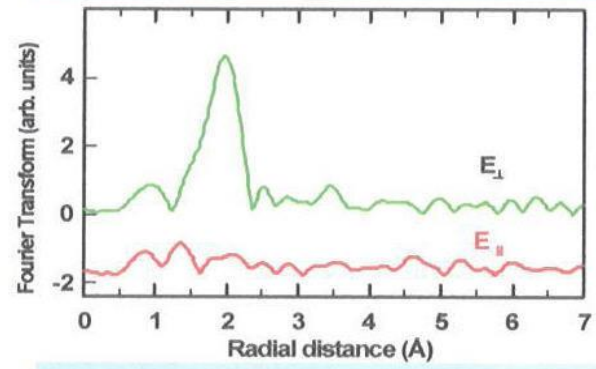
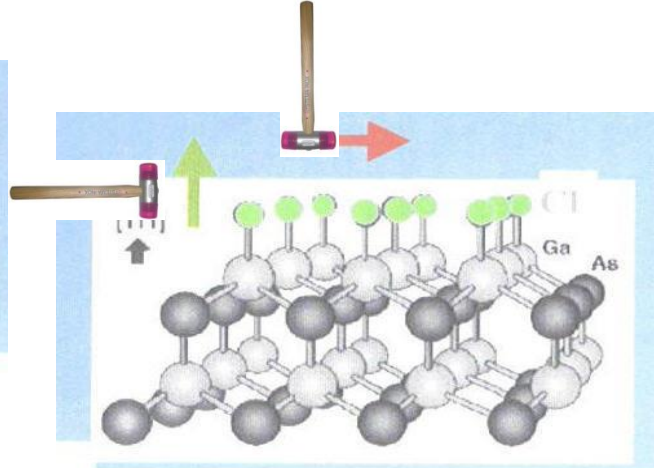
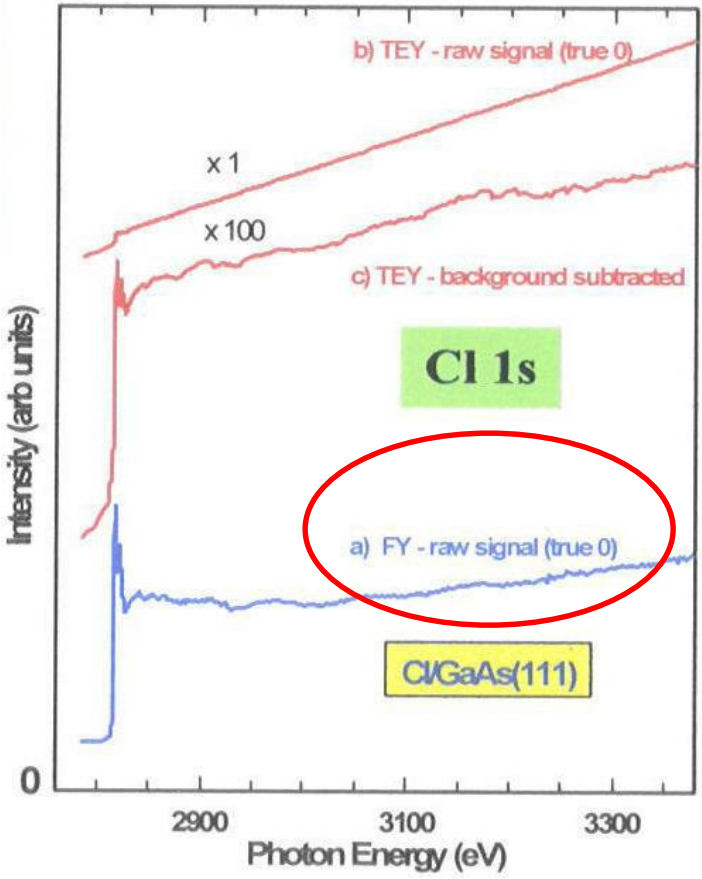


Metals in mouse kidney tissue (hard X-ray)

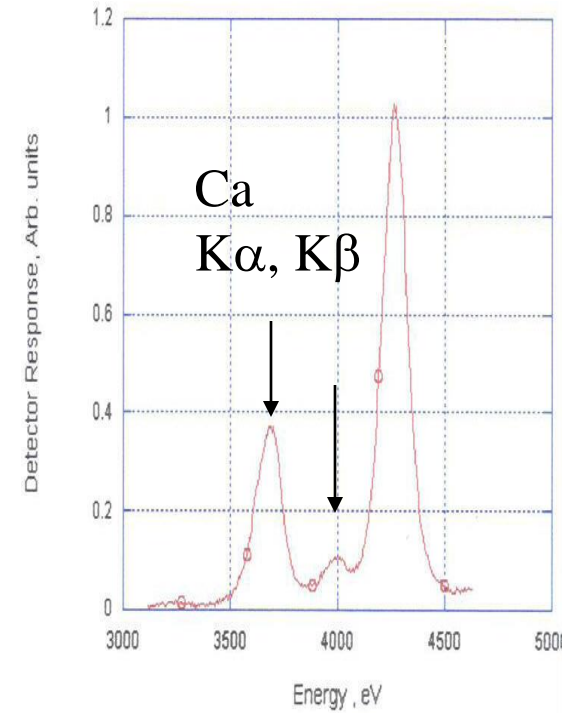
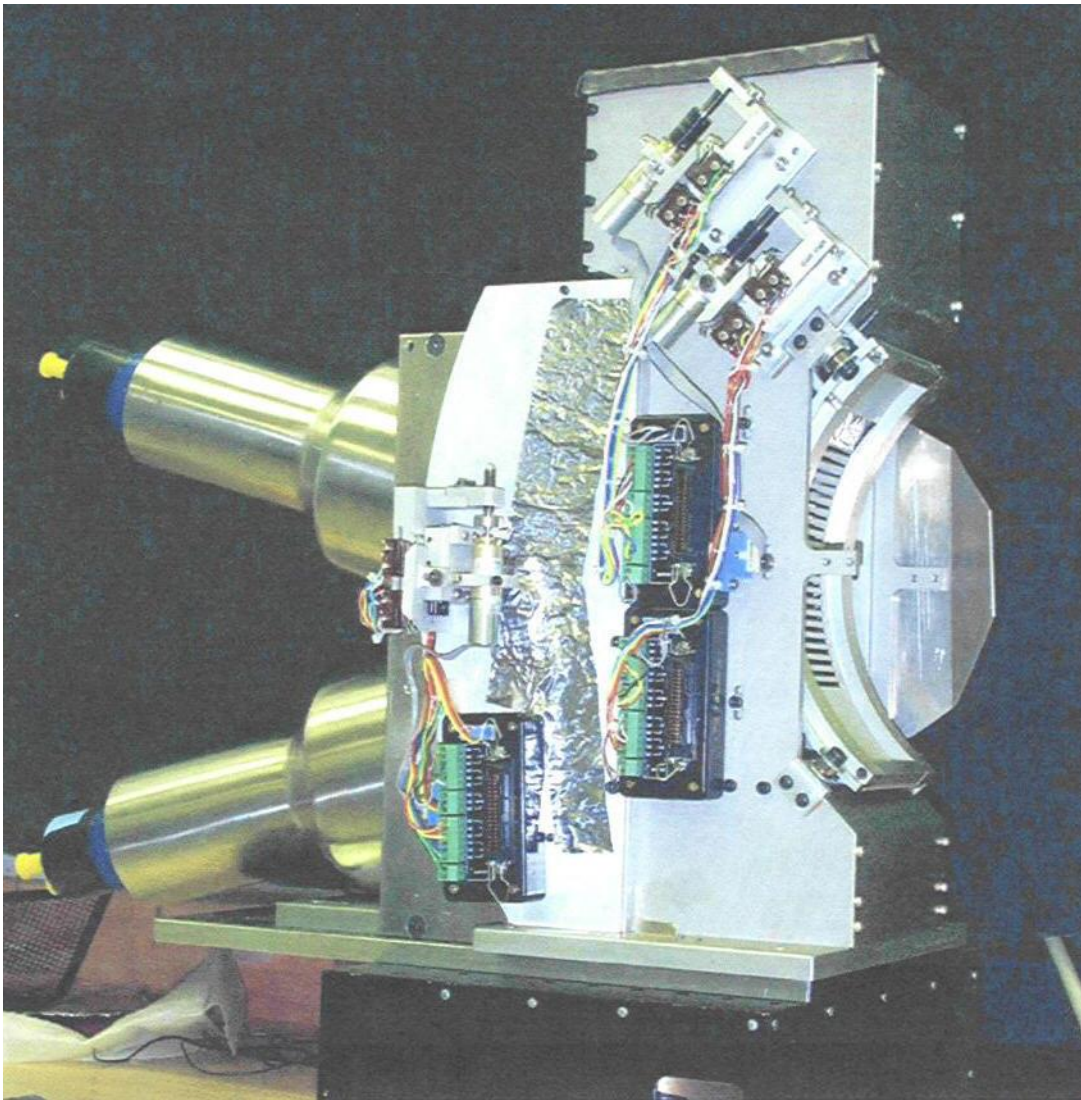
Incident X-ray
 $h\nu = 10 \text{ keV}$



Cl / GaAs(111)
Polarization dependent
X-ray absorption (EXAFS)



Lu, soft x-ray [Phys. Rev. B, 58 (1998)]



Energy scan: elastic,
Ca K α and K β
 $\Delta E = 150$ eV at Ca K α

20 element Multi-Array Analyzer Detector (MAAD)
[Ke Zhang et al. HD Technologies Inc.]

Variable width

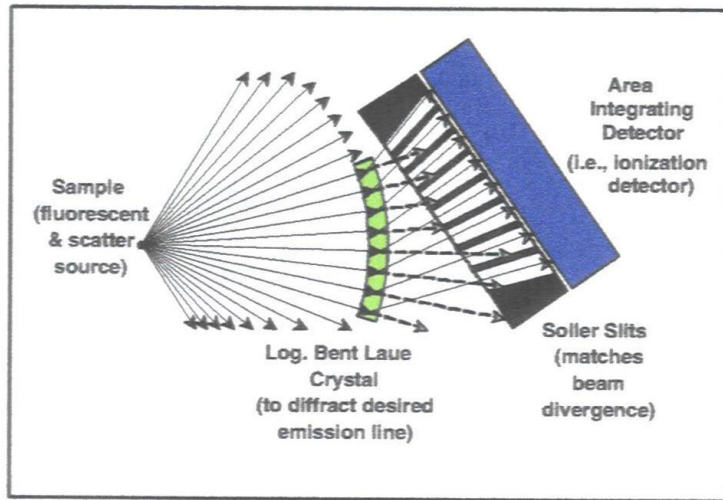


FIG. 1. A sketch of the "logarithmic bent Laue analyzer" concept.

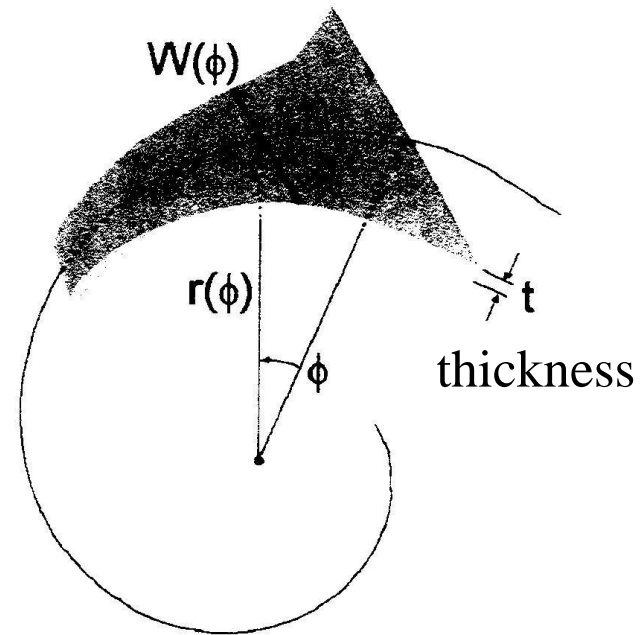


FIG. 2. Crystal bent to log spiral shape.

Schematic of an asymmetric cut Si (100) wafer where in polar Coordinate: $r(\phi) = ae^{b\phi}$,
 $b = \tan \theta$ [Khelashvili et al. Rev. Sci. Instru. 73, 1534 (2002)]

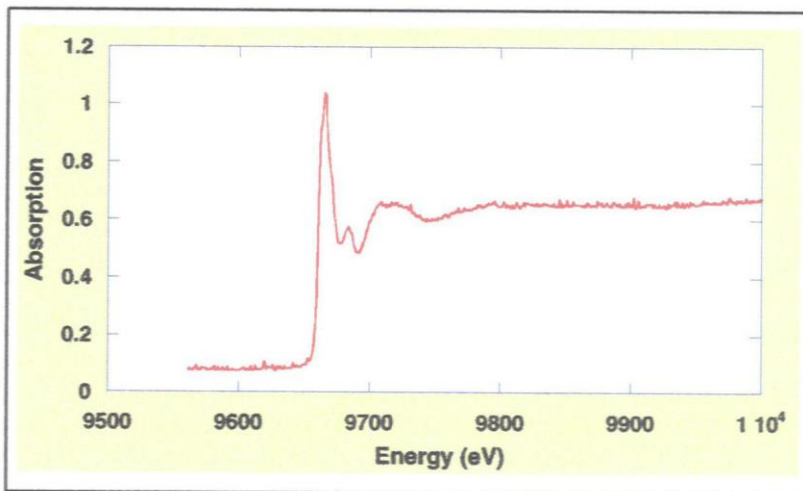


FIG. 2. An XAFS spectrum of around 65 ppm Zn - adsorbed to montmorillonite clay - measured with Logarithmic Bent Laue Analyzer.

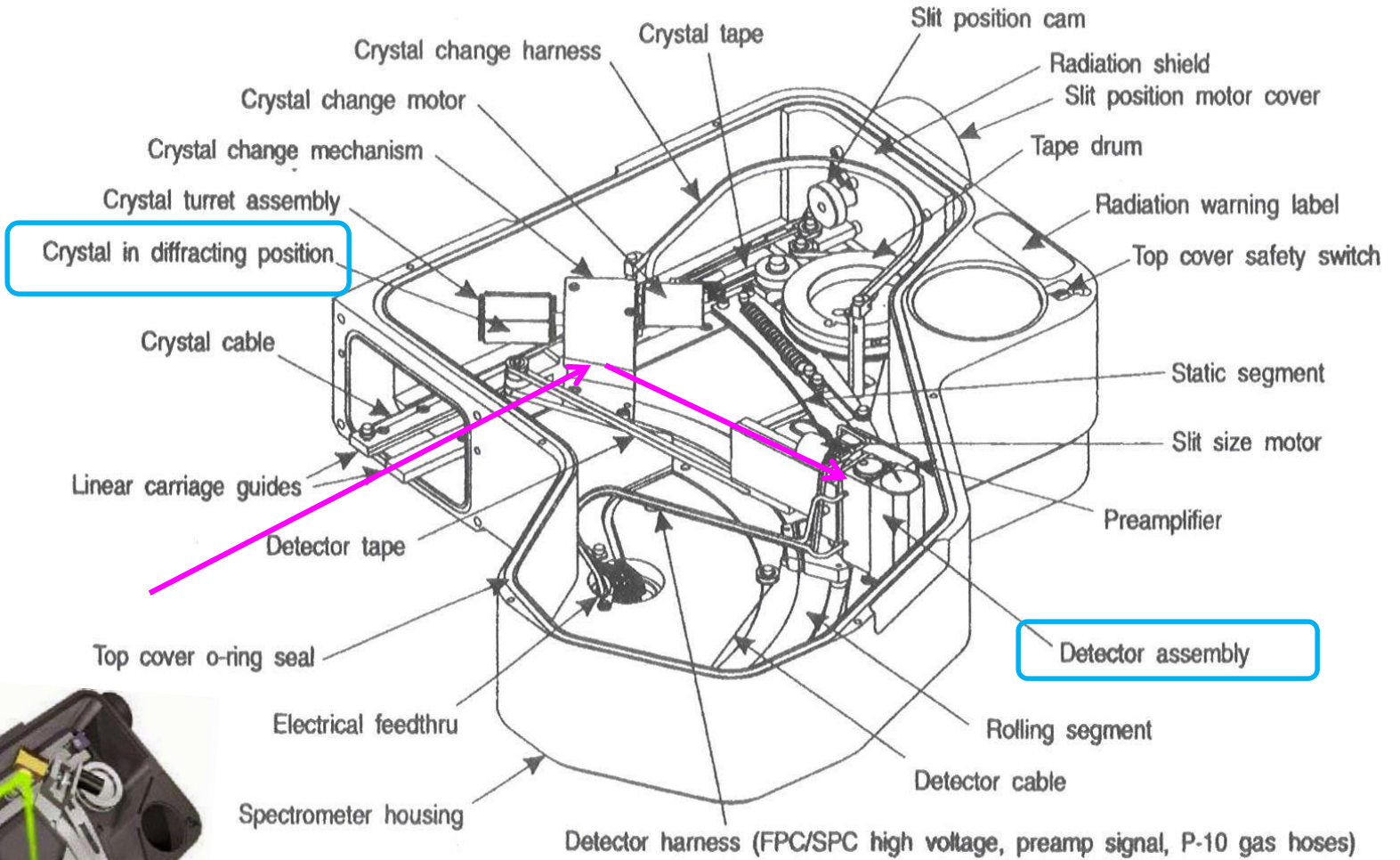
Detection Schemes (continues..)

- Low sensitivity, good resolution WDX
(Rowland circle crystal optics)

LiF crystals (electron microprobes): 2-5 keV
[e.g. PNC-CAT]

Ge (3,3,3), Si (4,4,0): Fluorescence > 5 keV

- Low sensitivity, good resolution (< 2 keV)
Grating monochromator [e.g. BL 8.0.1, ALS]



WDX from an electron microprobe

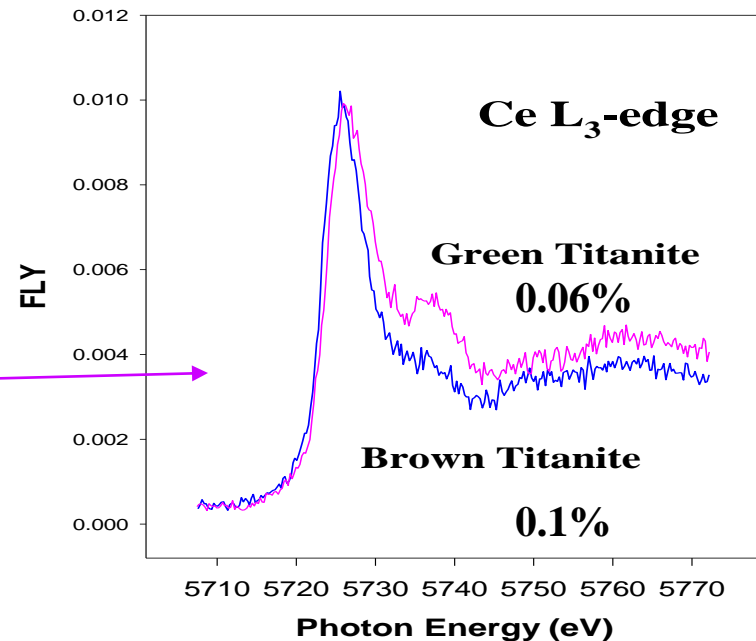
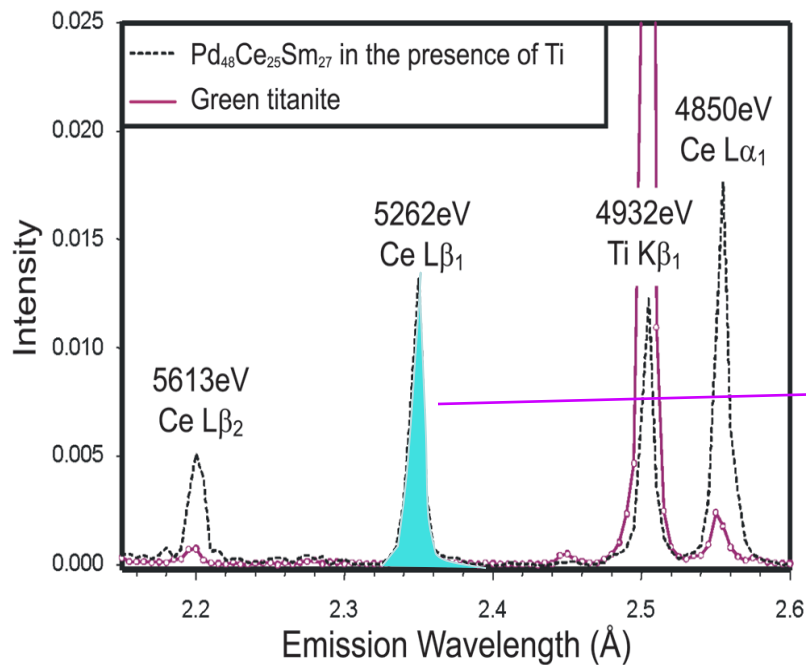
Ce L₃ edge of Ce in Titanite: WDX detection

Ti K β_1 4932 eV

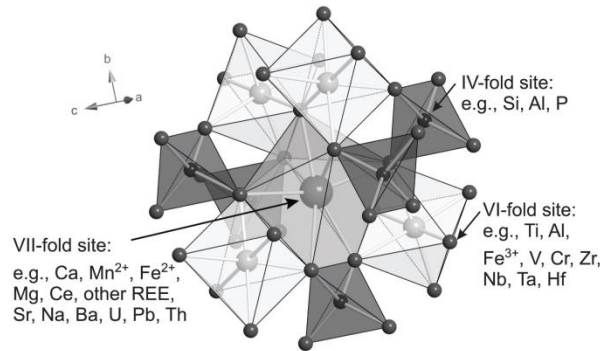
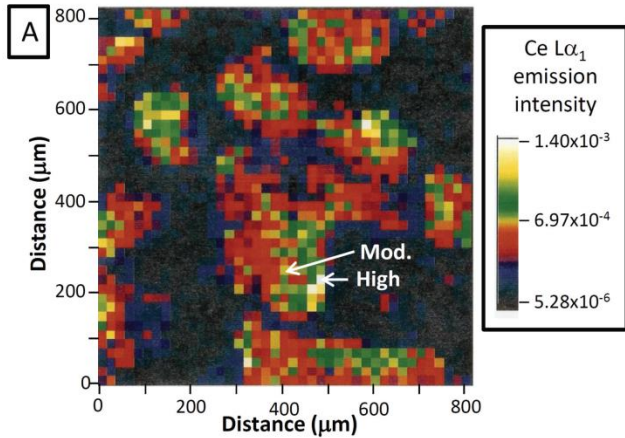
Ce L α_2 4823.0 eV

Ce L α_1 4840.8 eV Ce L β_1 5262.2 eV

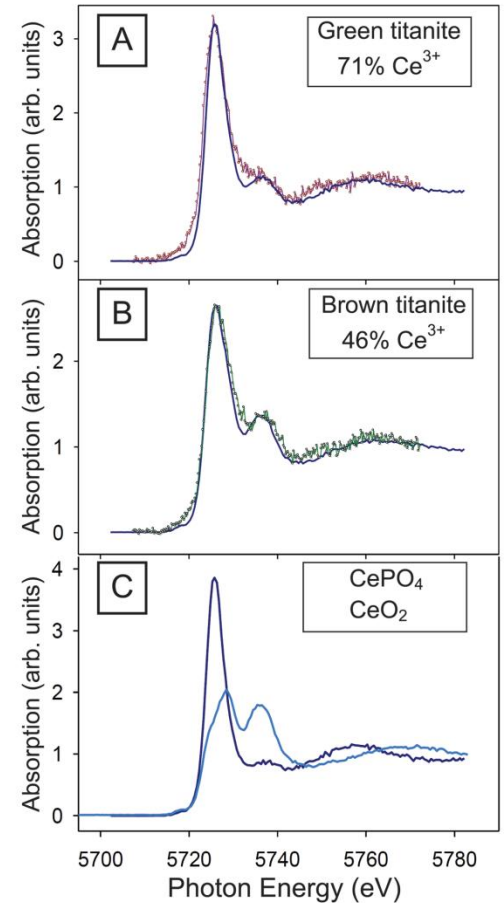
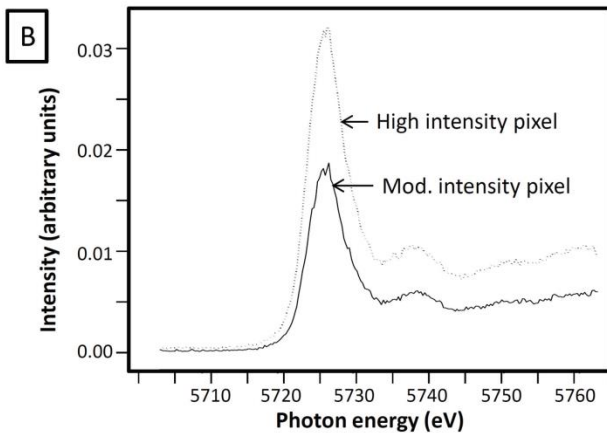
Neither SS nor WDX detector will resolve Ce L α from Ti K β
But WDX can resolve Ce L β nicely



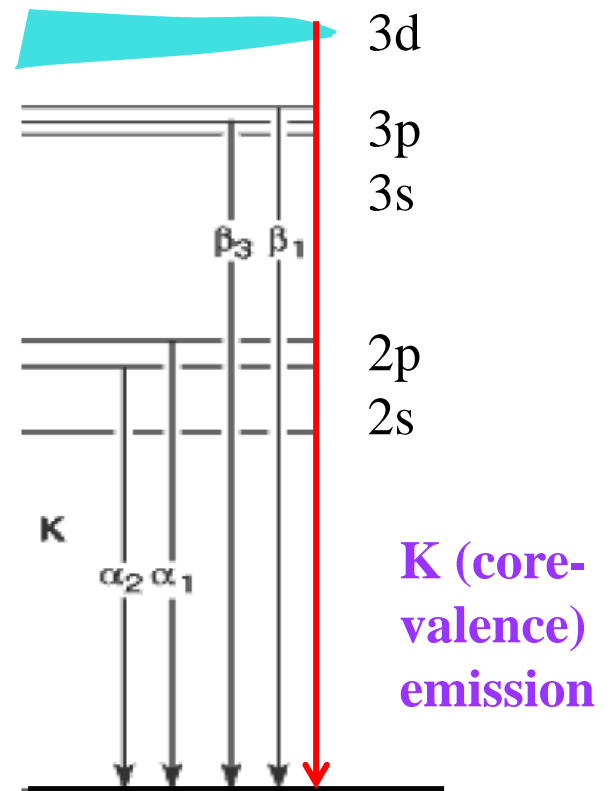
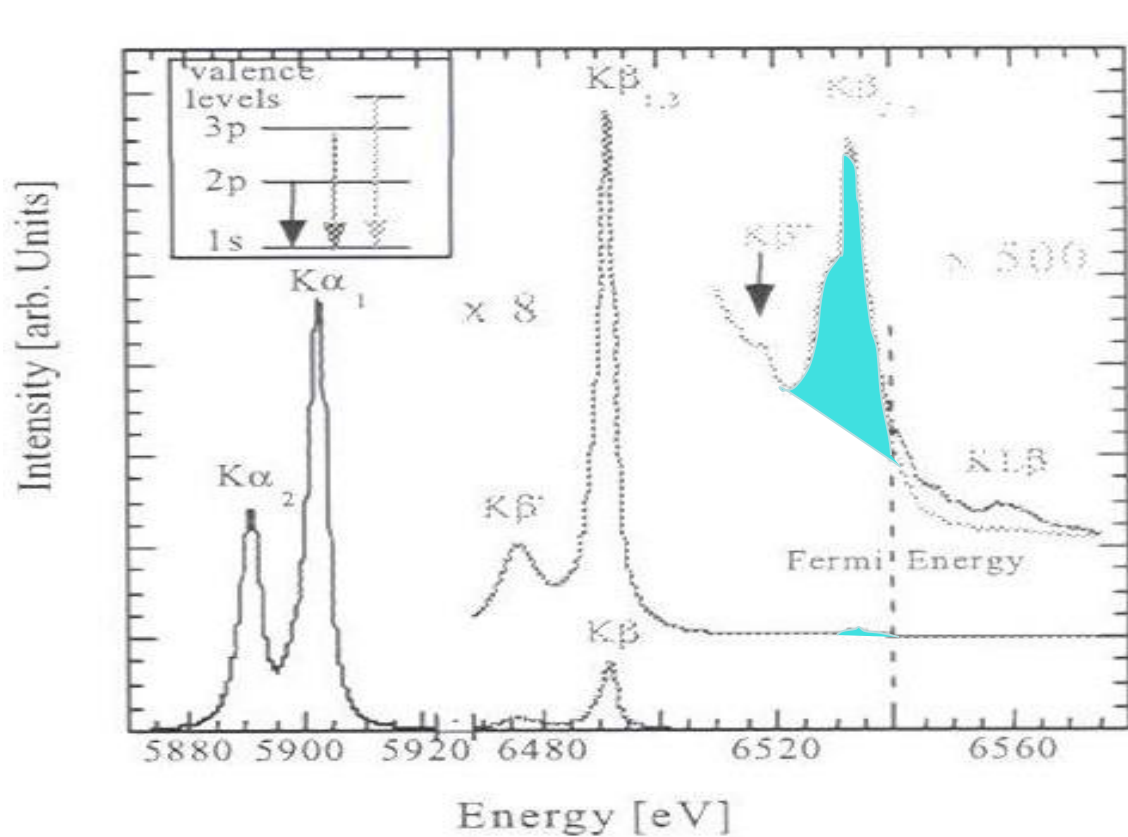
Ce in Titanite: Microprobe using WDX



Titanite (CaTiSiO_5 , sphene) is a common mineral in mafic-felsic igneous and metamorphic rocks, and it is widely used for geochronologic and petrogenetic studies



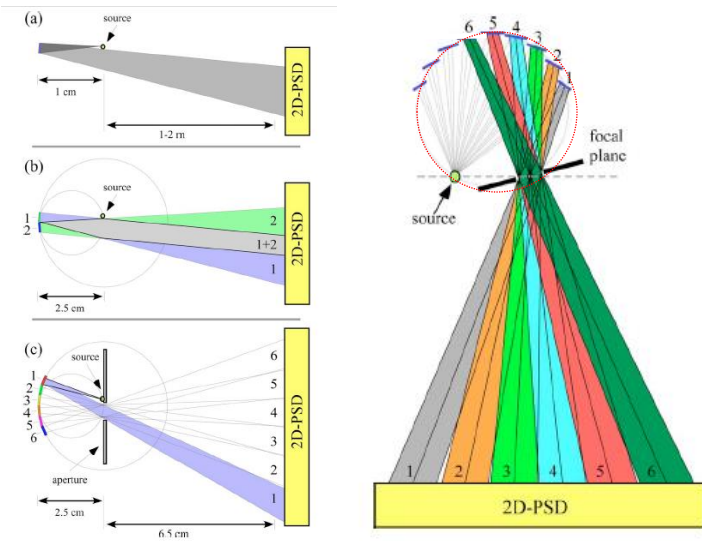
American Mineralogist, Volume 98, 110–119, (2013)



K Fluorescence of MnO using Ge(333) and Si(440) crystals (Rowland circle optics, S. Cramer X-25 NSLS, 2002)

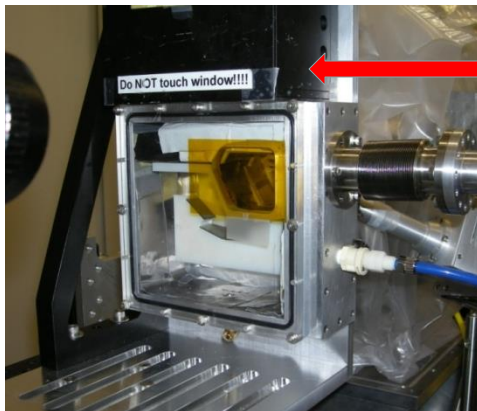
High resolution fluorescence using Minixs

Rowland Circle



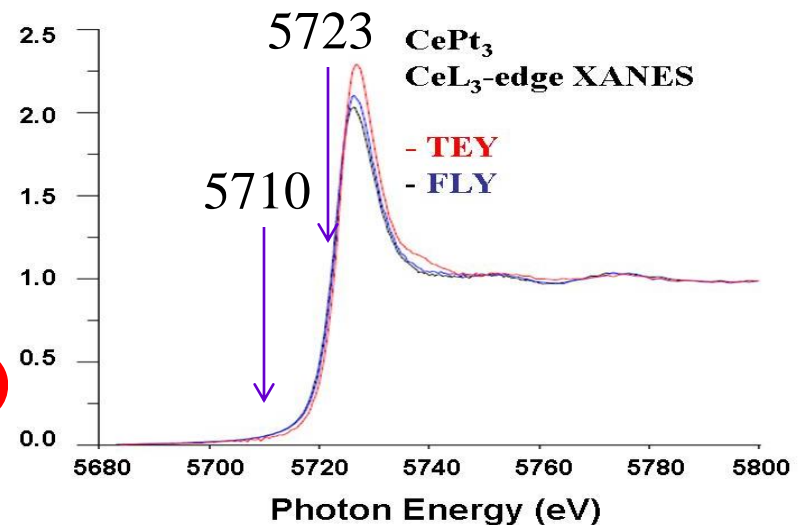
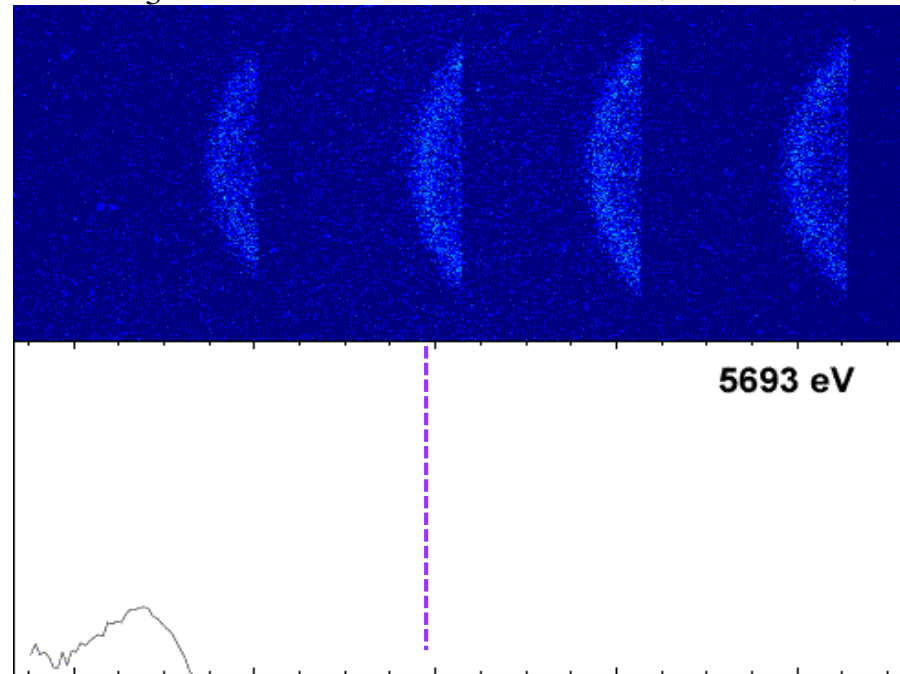
Dickinson et al. *Rev. Sci. Instrum.* **79**, 123112 (2008)

Pilatus 100K PAD



Ce L_{α} Ge(440)

CePd₃ RXES data collection (~3 hours)

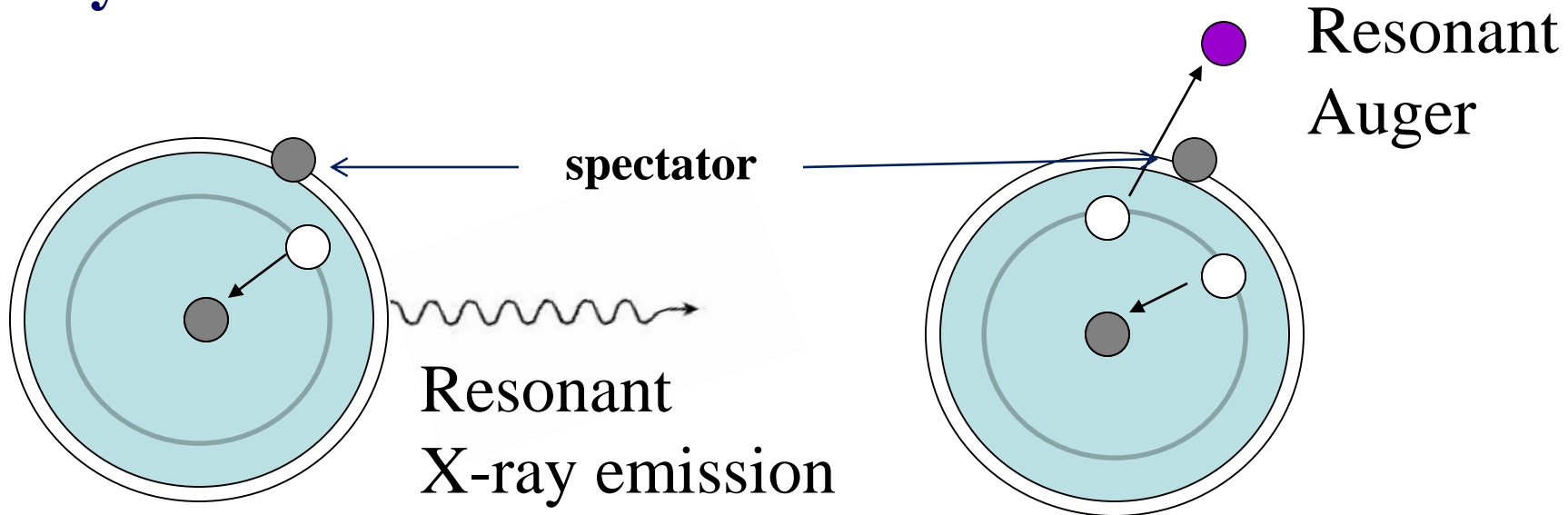


Absorption & de-excitation @ resonance:

Unoccupied bound state

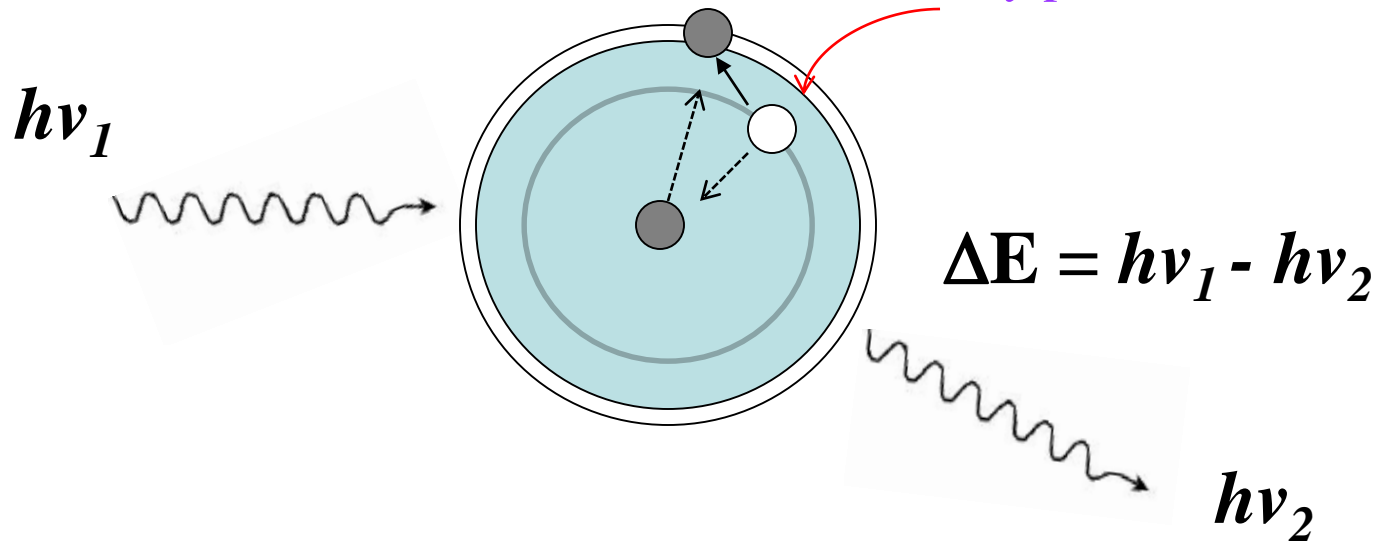
Photon channel:
X-ray fluorescence

Electron channel:
Auger electron



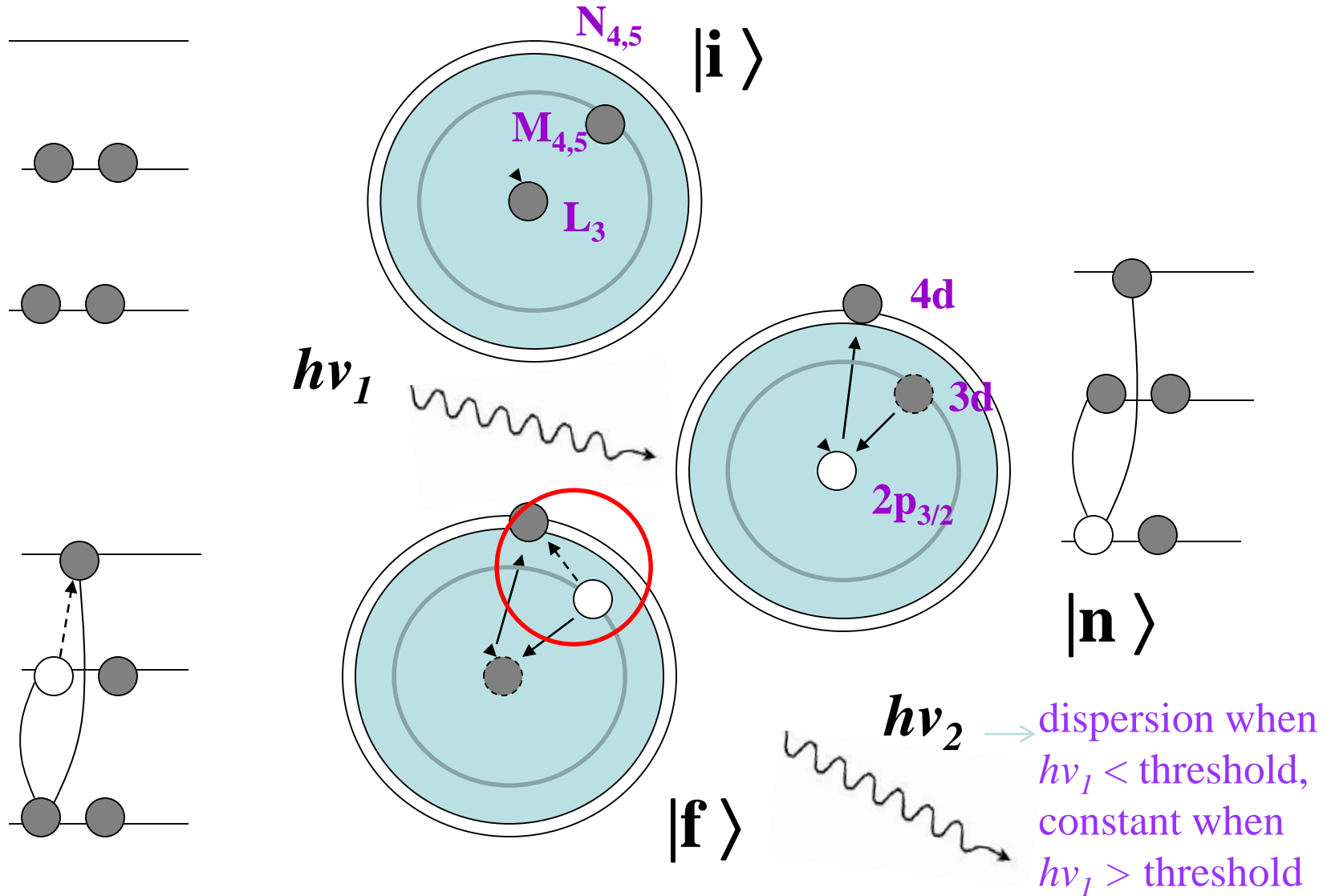
Inelastic X-ray scattering (core level excitation)

Same shallow hole left behind when the e drops down to fill a deeper core hole emitting a fluorescence X-ray photon



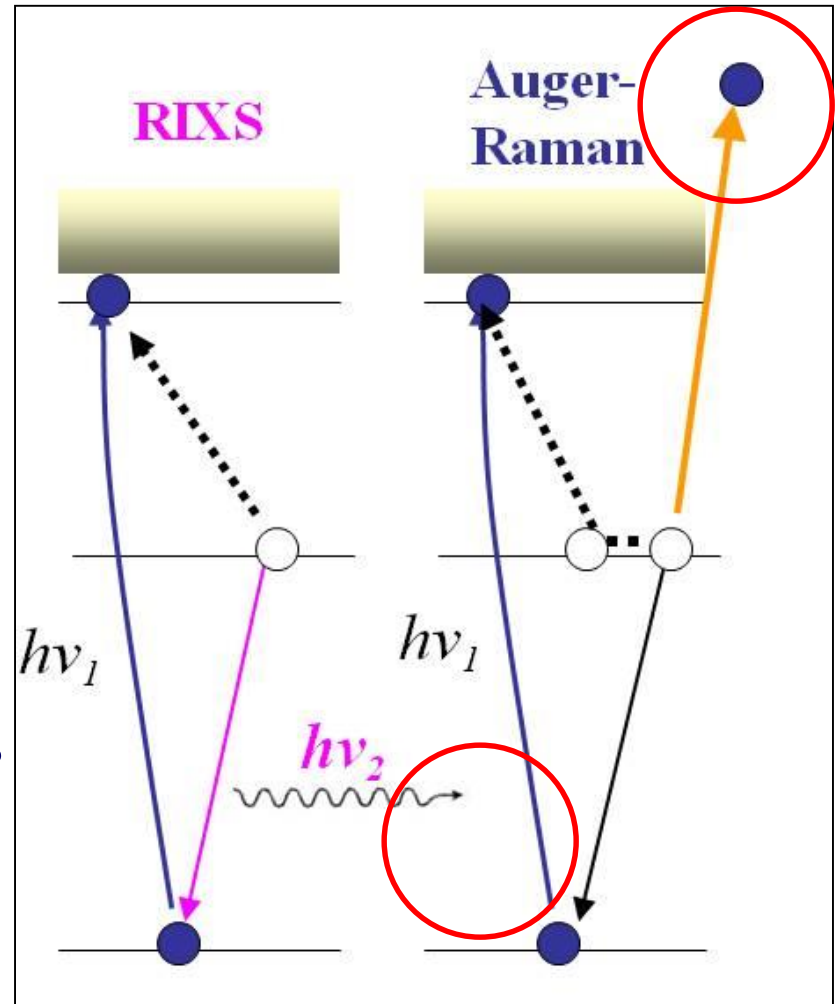
Cross-section is very small except at resonance
When $h\nu_1 = h\nu$ (threshold resonance)

RIXS @ Ce L₃-edge

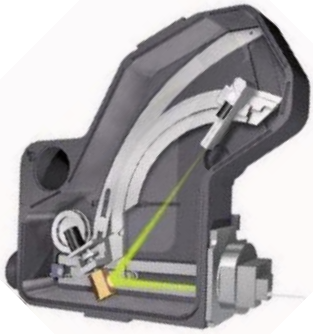


RIXS measurements (electron and photon)

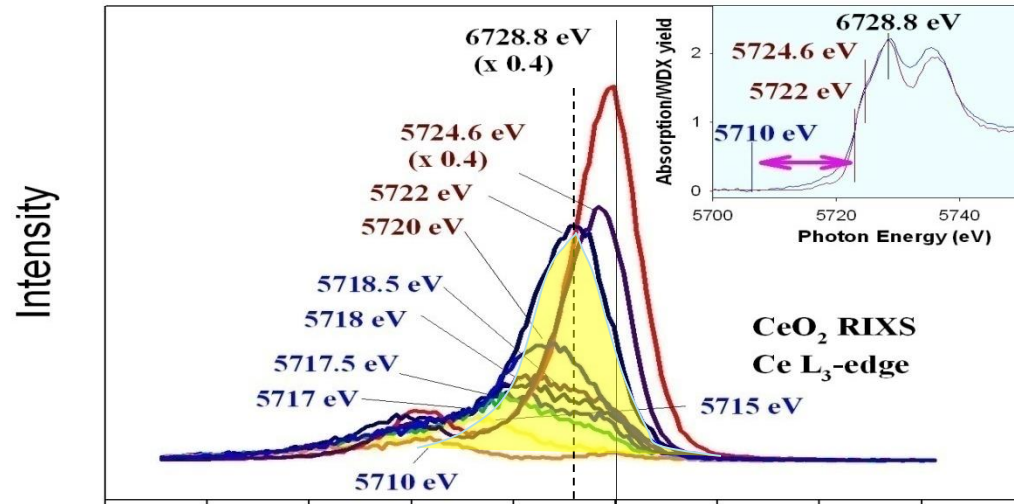
- Experiment requires **high flux, high photon energy resolution** (incident and emission) and **high electron energy resolution**
- Information: **dispersion** (constant energy transfer); **core-hole lifetime suppression**



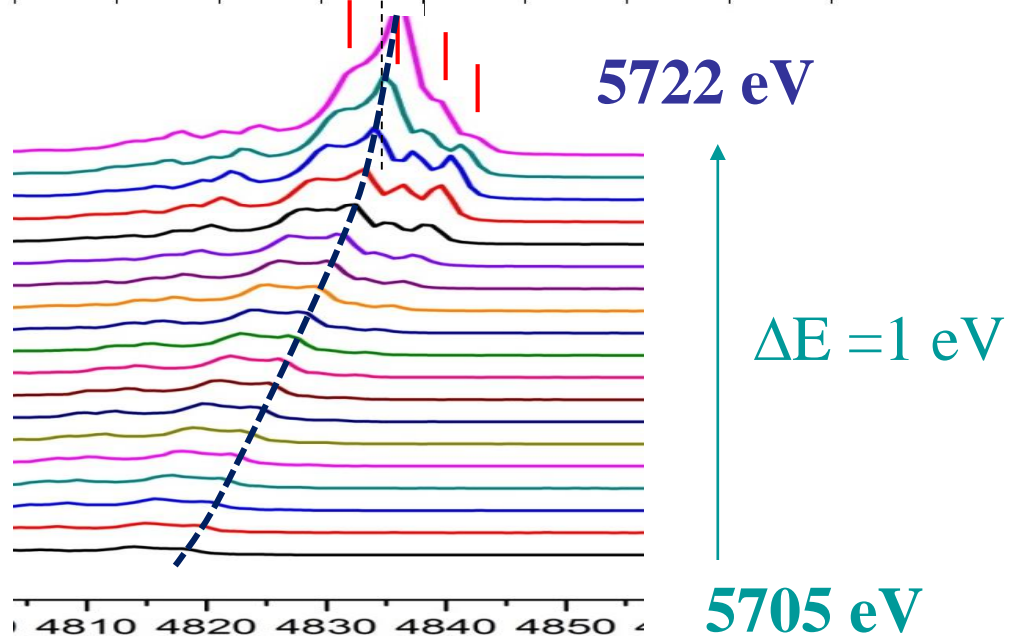
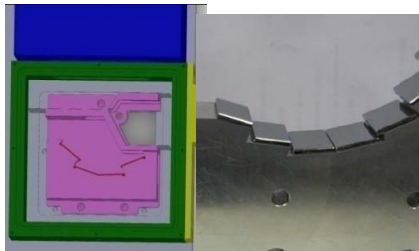
Resonant Inelastic X-ray Scattering (RIXS)



WDX
 $\Delta E \sim 6 \text{ eV}$



MiniXS
 $\Delta E < 1 \text{ eV}$

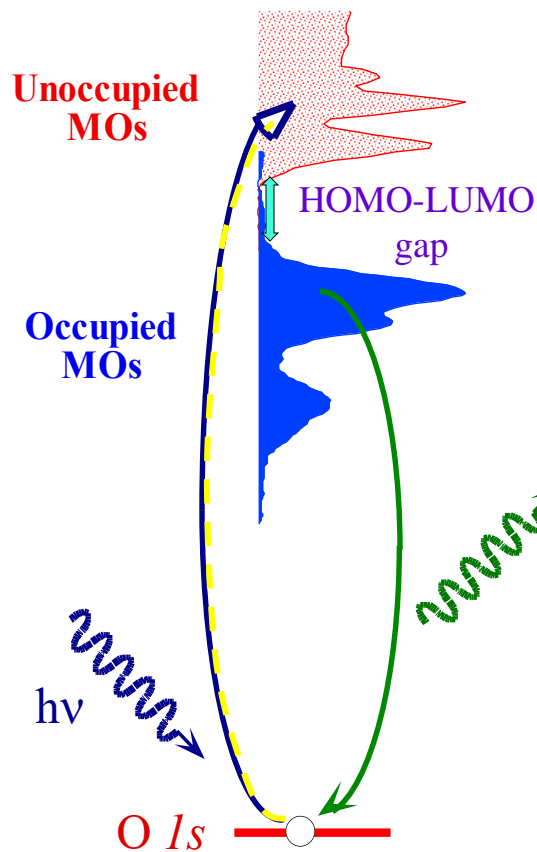


XES : X-ray Emission Spectroscopy

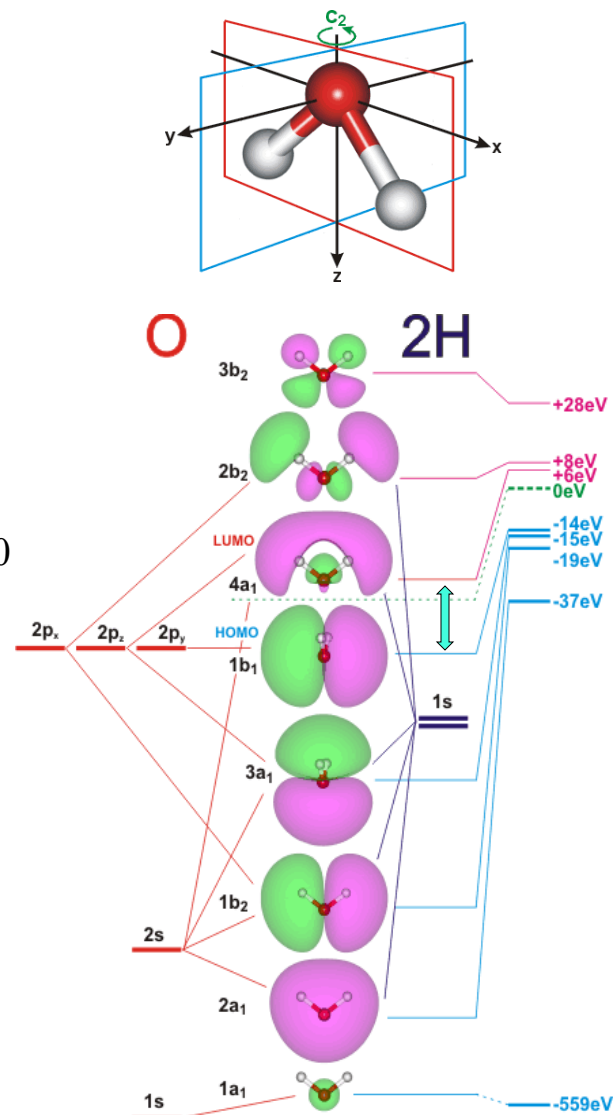
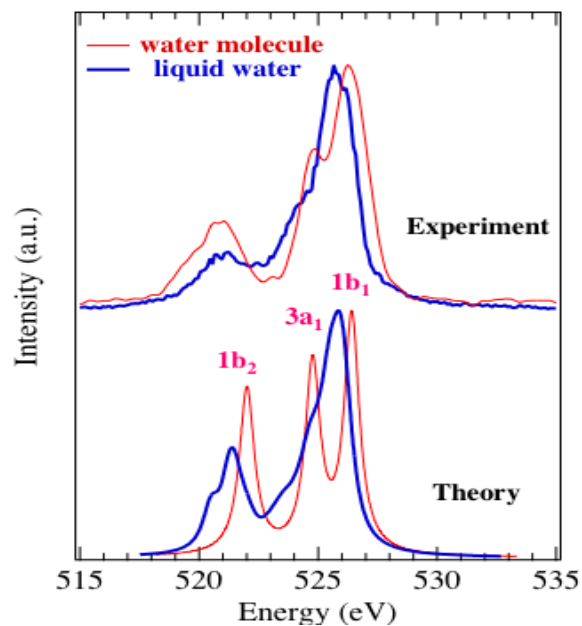
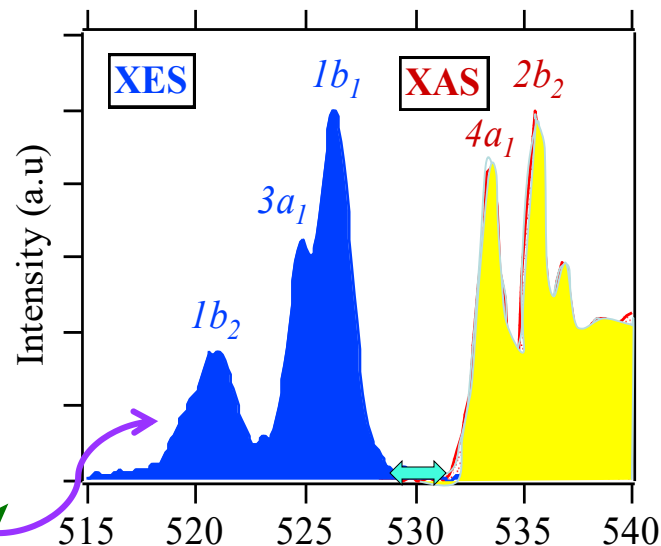
X -ray photons in, x-ray photons out

For shallow core excitations, XES measures the projected densities of states of the element in the valence band → **element and chemical specificity**

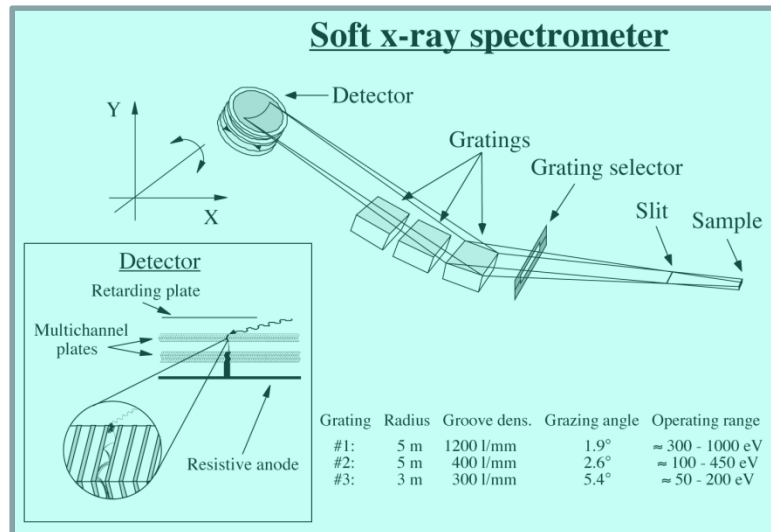
O K-edge XAS and XES of H₂O



Guo et al., PRL 89,
137402 (2002)



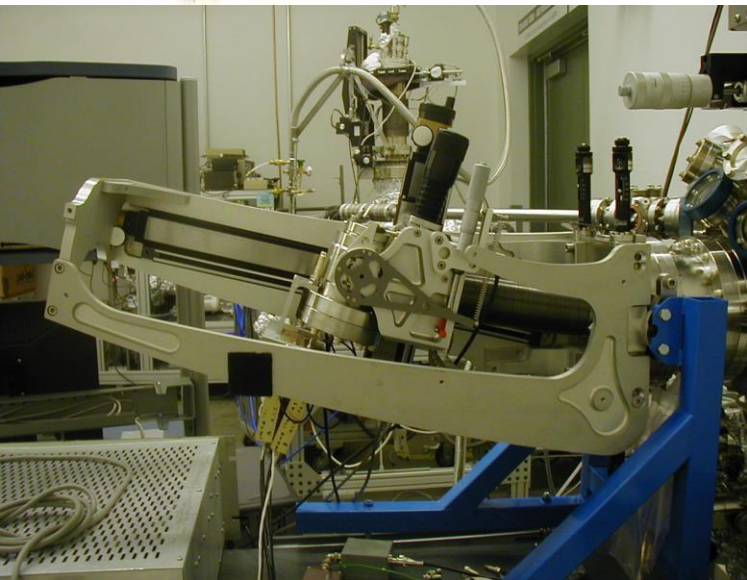
Resolution of Grating Spectrometer



1200 lines/mm, 5 m, resolution:
 670 meV (at 1000 eV)
 370 meV (O K-edge at 525 eV)

400 lines/mm, 5 m, resolution:
 300 meV (C K-edge at 275 eV)
 100 meV (S L-edge at 150 eV)

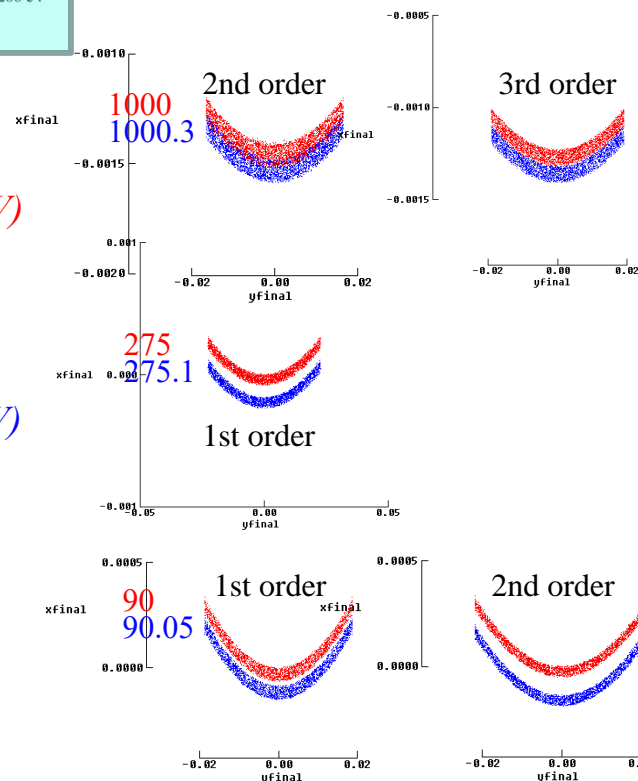
300 lines/mm, 3 m, resolution:
 80 meV (Cu M-edge at 72 eV)



1800 lines/mm, 5 m, resolution:
 300 meV (at 1000 eV)
 150 meV (O K-edge at 525 eV)

1200 lines/mm, 5 m, resolution:
 100 meV (C K-edge at 275 eV)

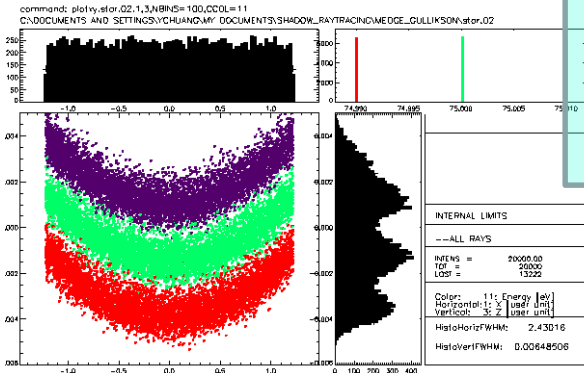
600 lines/mm, 3 m, resolution:
 50 meV at 90 eV



Slit size 10um, Detector resolution 50um

Very High Resolution Spectrometer

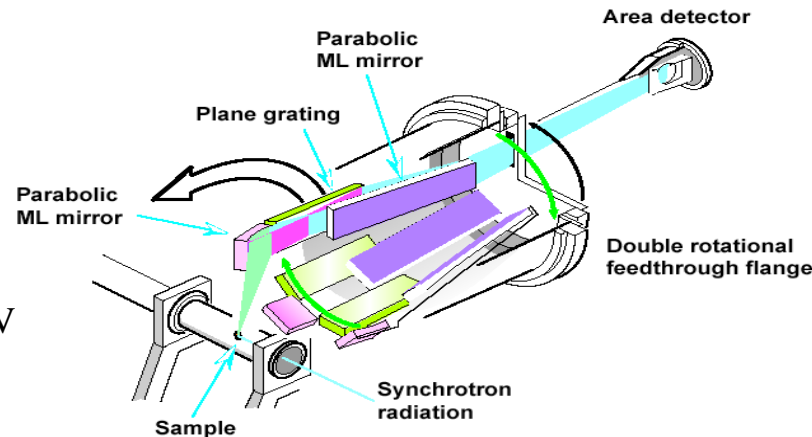
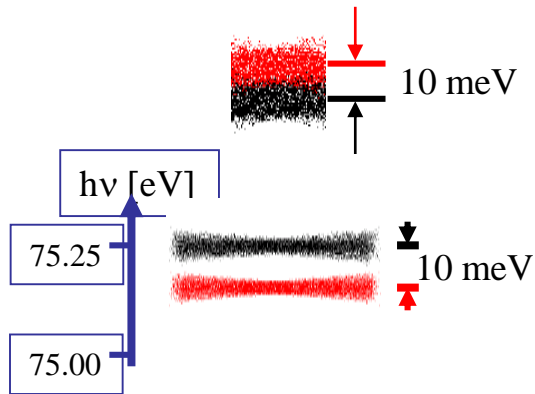
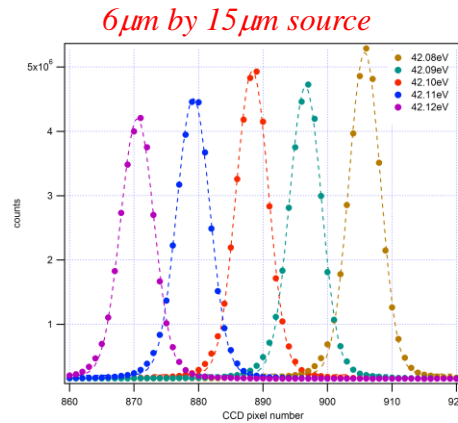
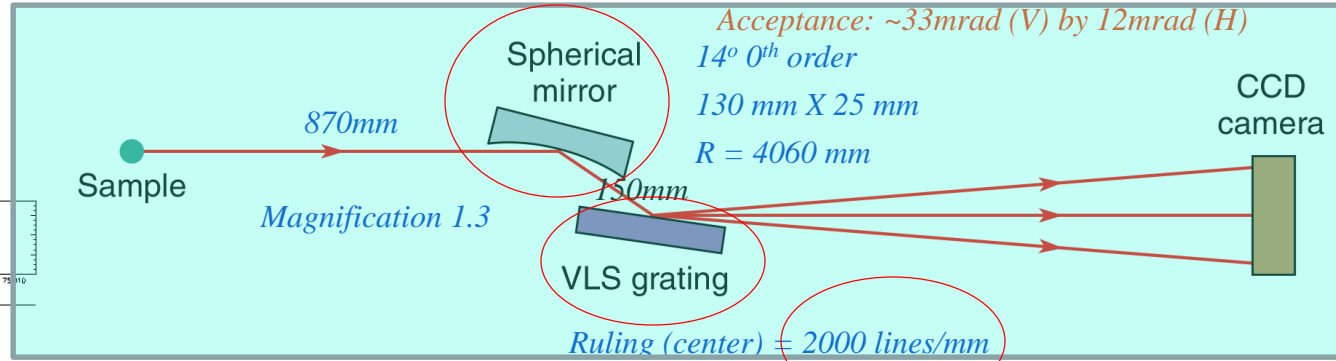
MERLIN at ALS:



*Cu 3p;
75eV +/- 10meV
RP ~ 7,500*

Courtesy of J.H. Guo

Uppsala University:



Source size: $6\ \mu\text{m} \times 60\ \mu\text{m}$
 Grating: 1200 l/mm
 Angle of incidence: 78 deg., outside order
 Acceptance angle: 100 mrad x 50 mrad

Source size: $6\ \mu\text{m} \times 60\ \mu\text{m}$
 Grating: 1800 l/mm
 Angle of incidence: 85 deg., inside order
 Acceptance angle: 20 mrad x 50 mrad

Intensity of X-ray emission Spectra

❖ Fluorescence Yield

- Photon hungry experiment
- **Resonance enhancement**

❖ Excitation

- Synchrotron radiation
- Undulator (Linear, EPU)
- Monochromator

❖ Detection

✓ *Diffraction efficiency of grating*

- *Blaze*
- *Grove density*
- *Surface quality*

✓ *Quantum efficiency of detector*

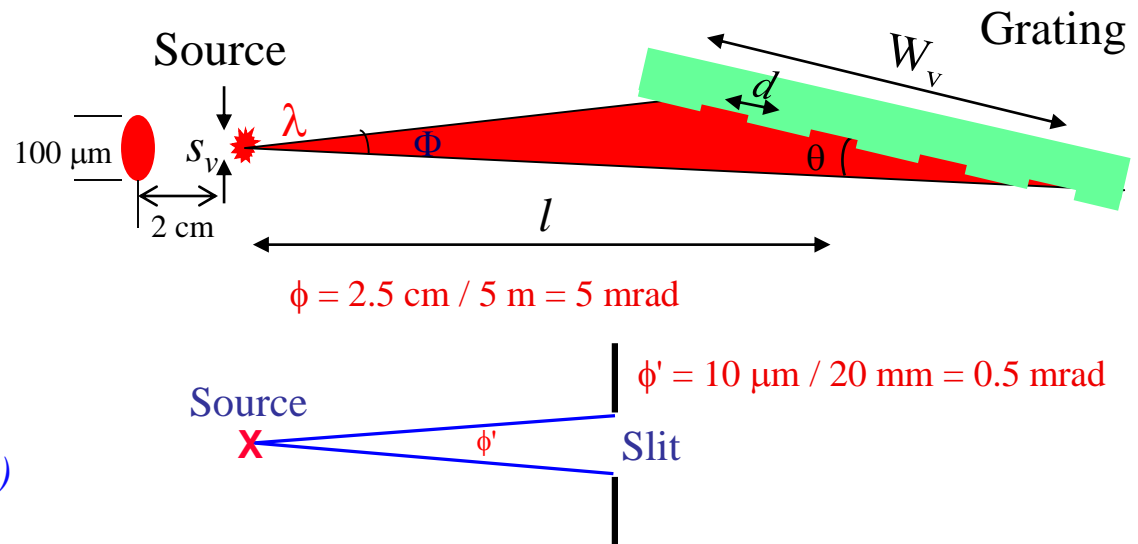
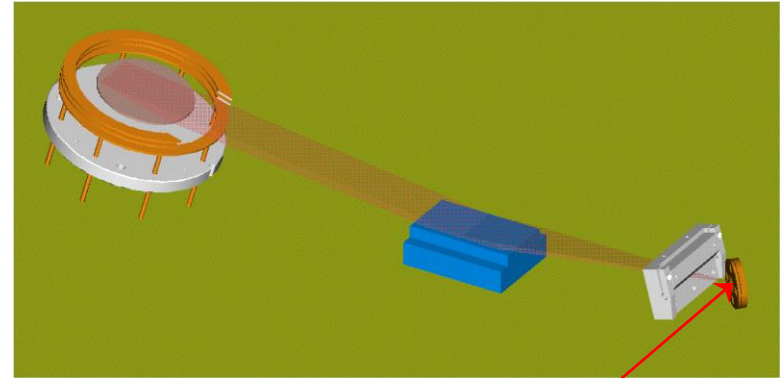
- *MCP (Photon cathode coating)*
- *CCD*

✓ *Beam spot size*

✓ *Solid angles to collected (slit size)*

Beam size (1994):

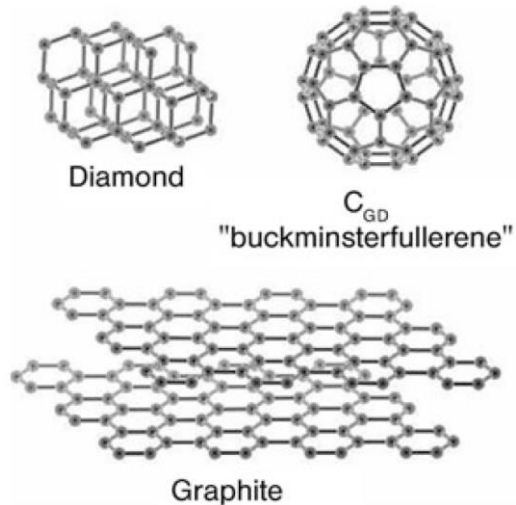
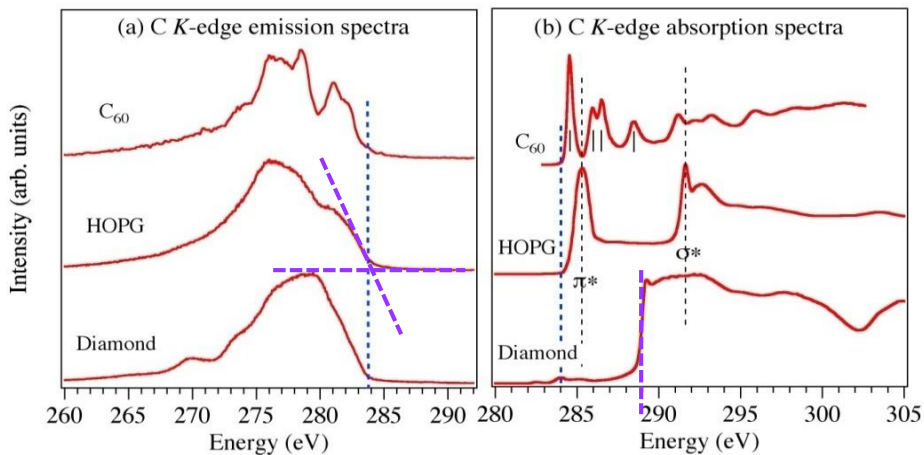
- 1 mm at BW3 (HASYLAB) & X1B (NSLS)
- 100 μm at BL7.0.1 (ALS)



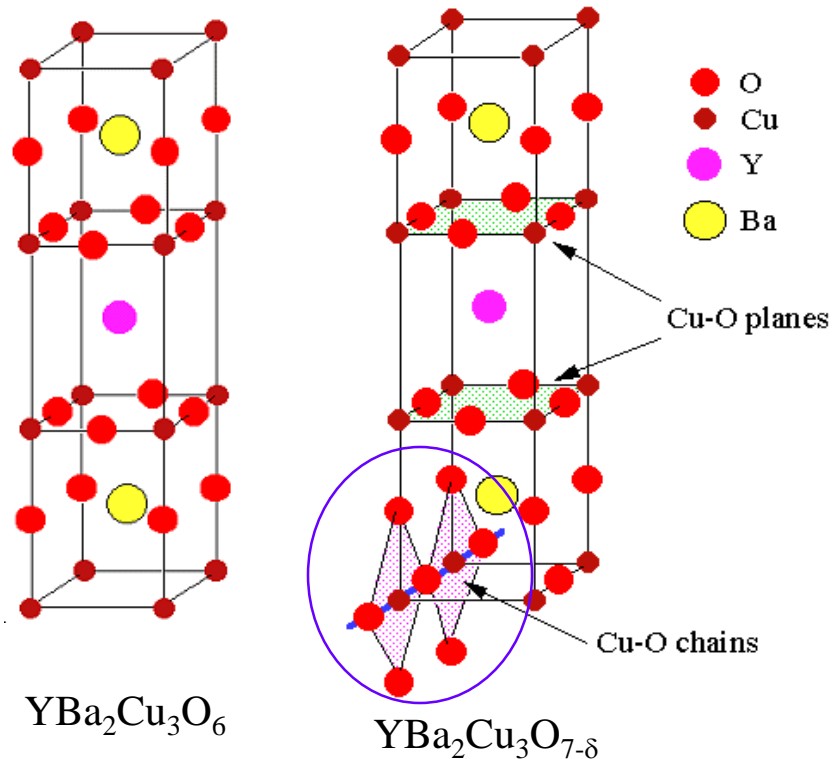
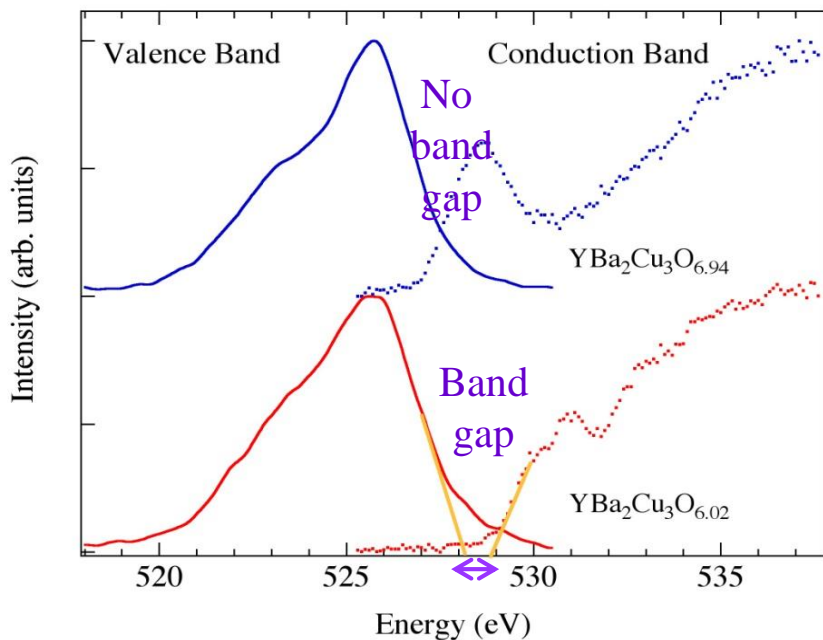
The beam size can be seen at 20 mm distance:

$$5 \text{ mrad} \times 20 \text{ mm} = 100 \mu\text{m}$$

XES + XAS : Bandgap determination



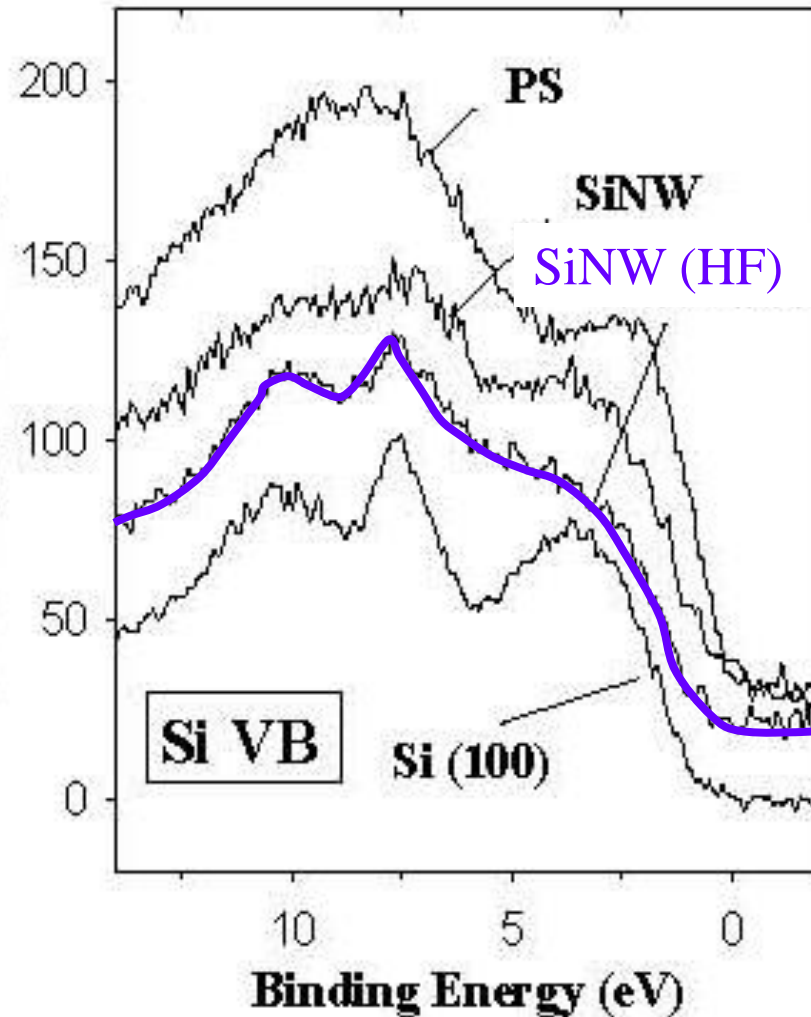
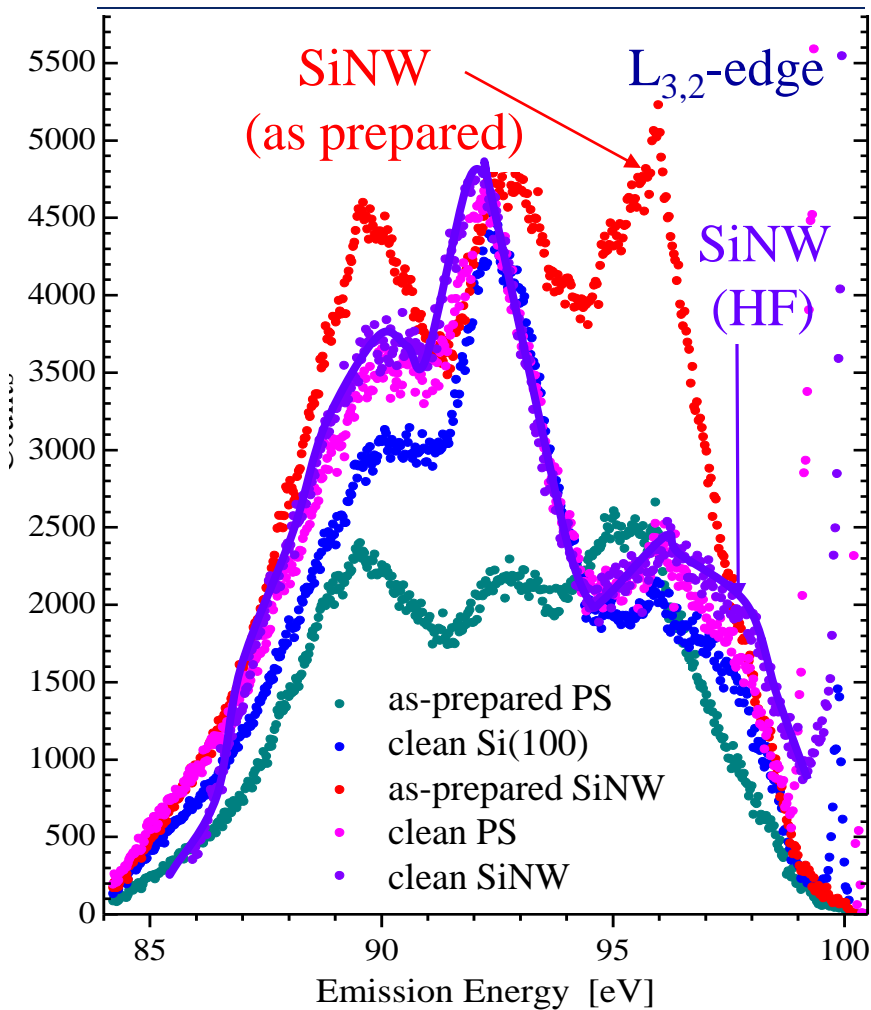
Guo, Int. J. Nanotech. 1-2, 193 (2004)



Guo et al., Phys. Rev. B 61, 9140 (2000)

Si L XES: Si \rightarrow Valence Band

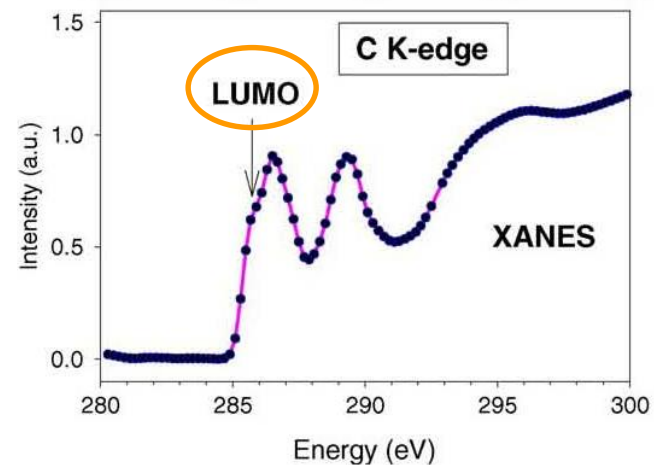
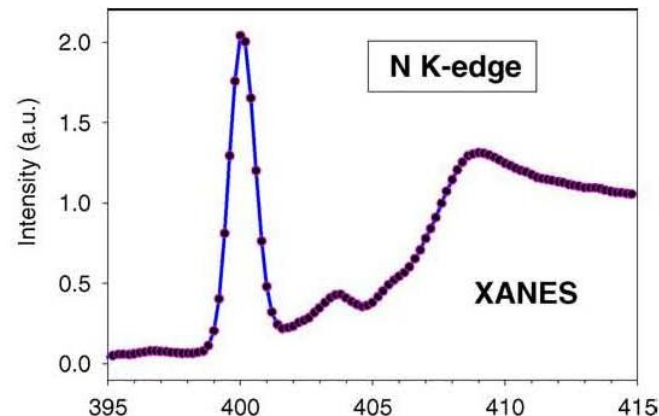
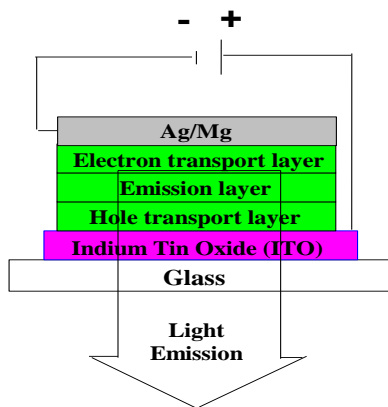
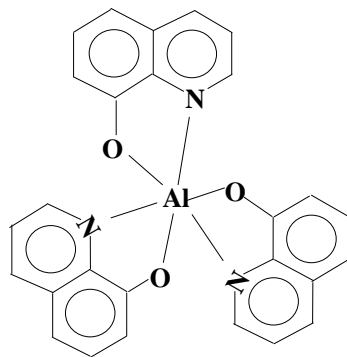
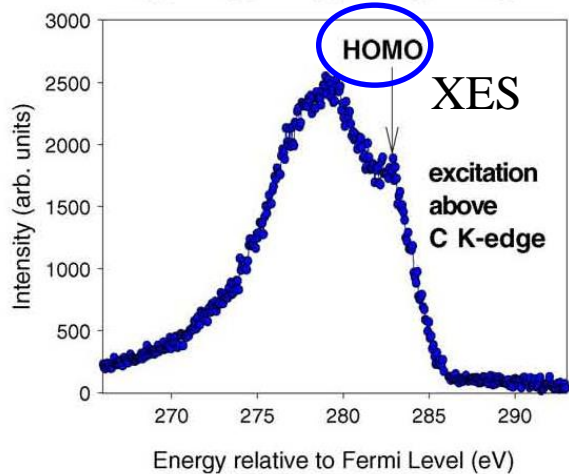
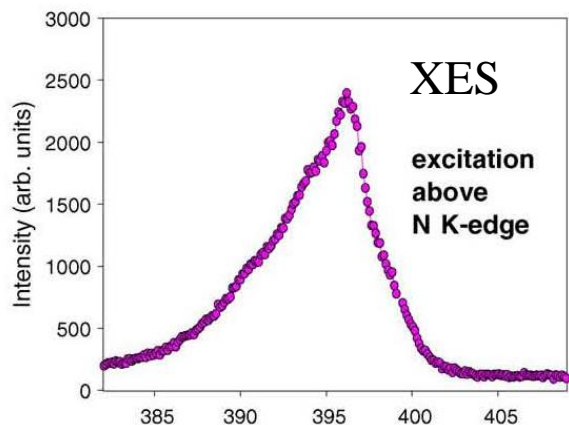
XPS: Valence Band



XES of Alq₃

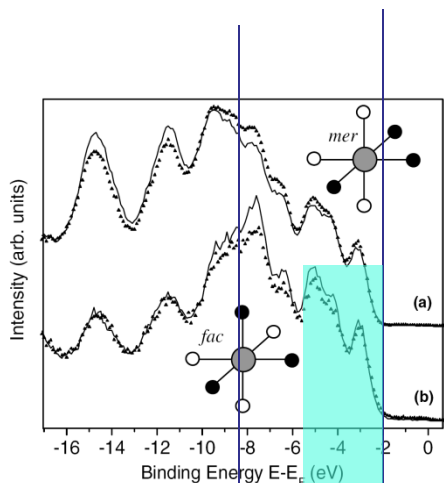
XES:PDOS (occupied)

XANES:PDOS (unoccupied)

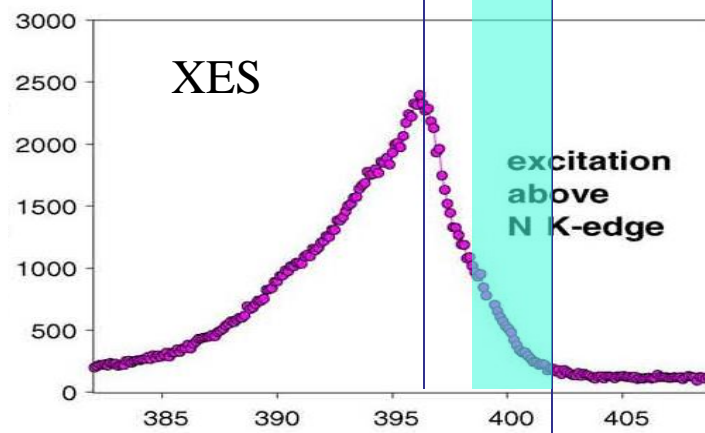
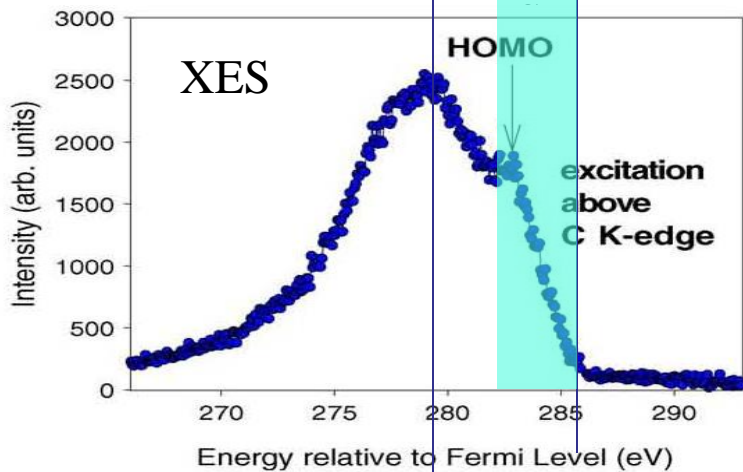
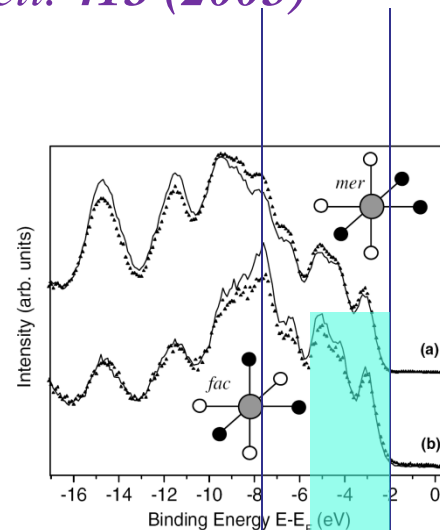


XES of Alq₃

XPS: Caruso et al. *Chem. Phys. Lett.* 413 (2005)



HOMO, HOMO-1:
Mainly C character



P.-S.G. Kim et. al. *J. Elect. Spectros.* , 901, 141(2005)

XES:PDOS (occupied)

XEOL: X-ray Excited Optical Luminescence

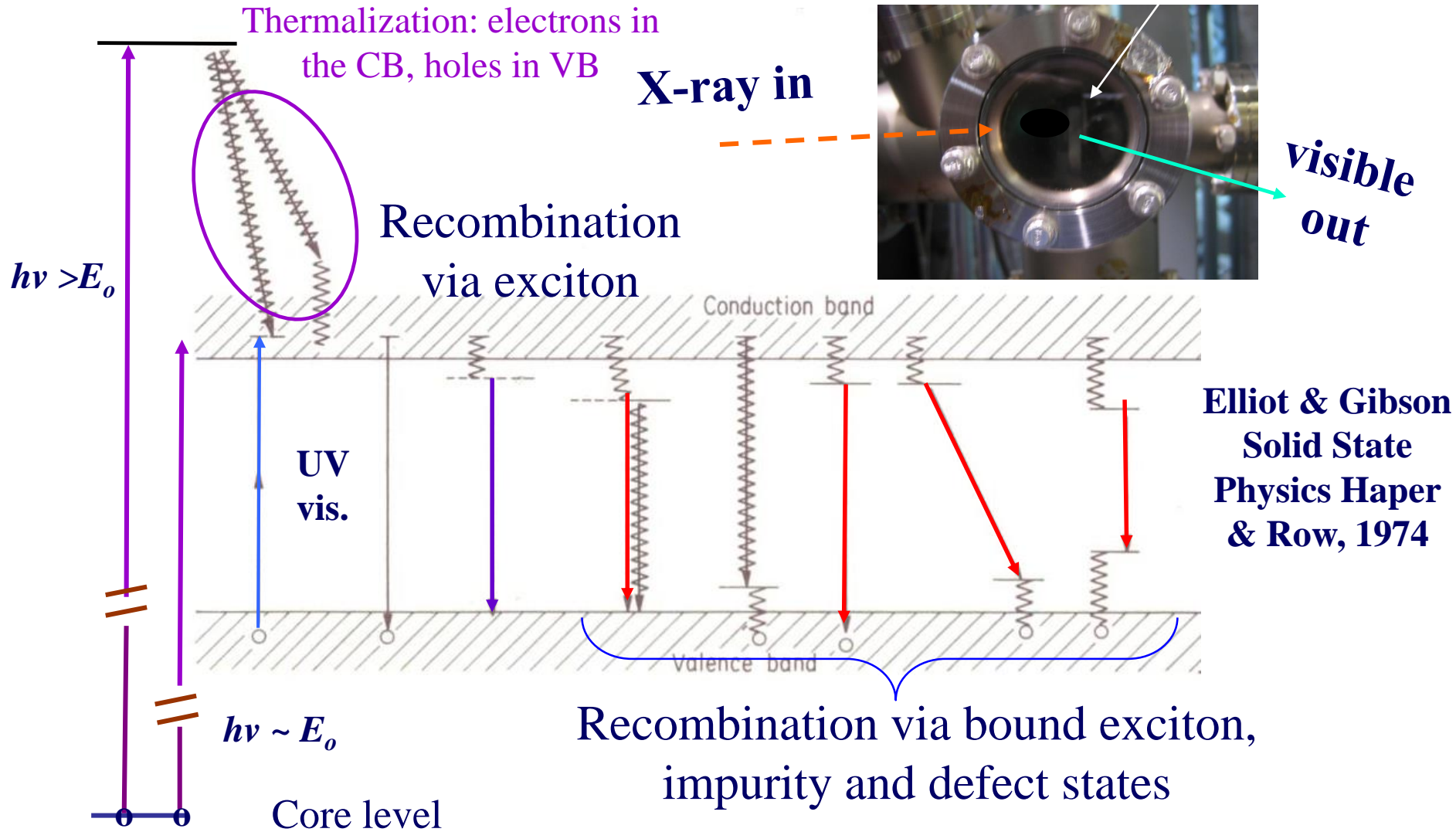
X-ray photons in, optical photons (UV, visible, near IR) out.

Luminescence can be element and excitation channel specific.

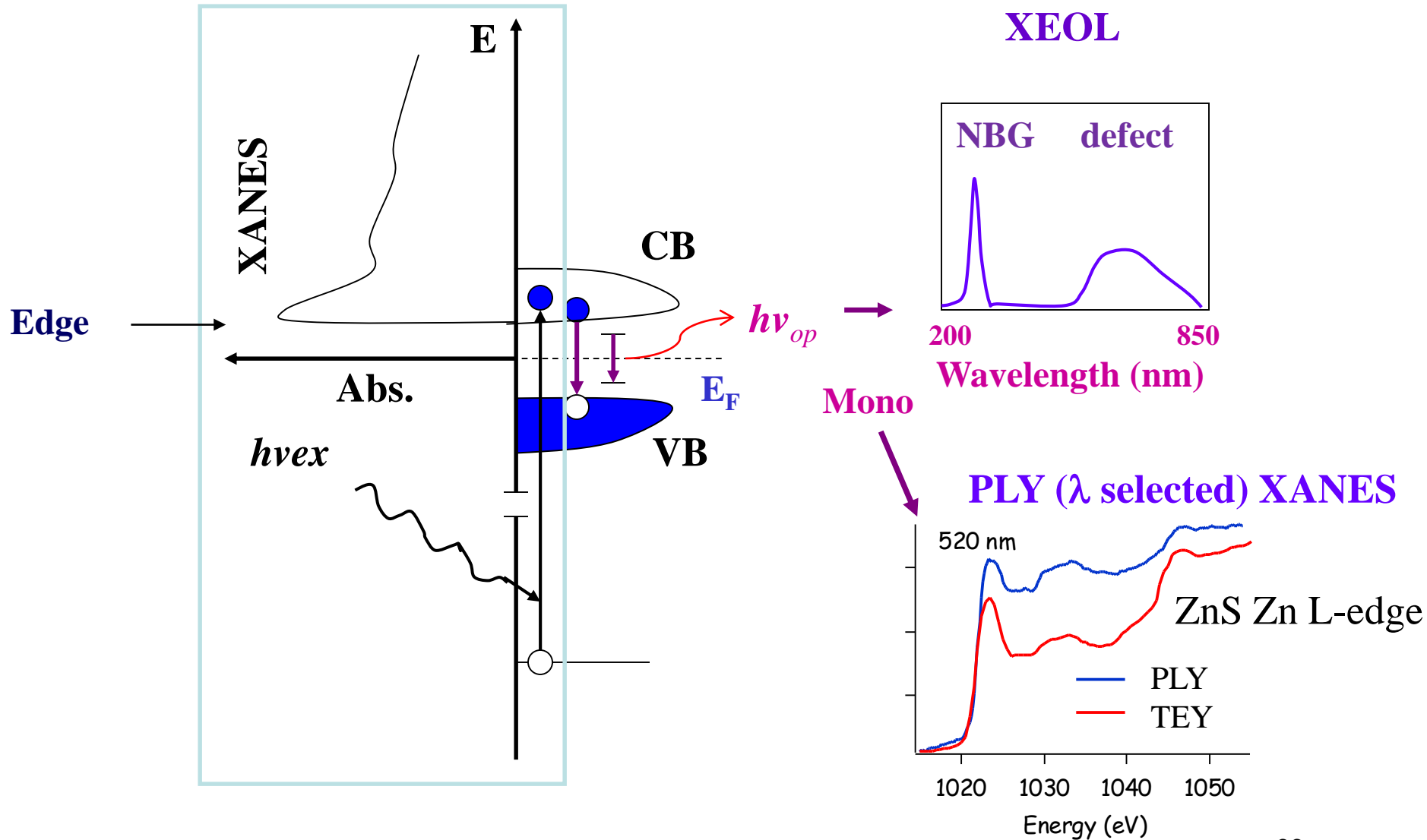
Monitor the absorption spectrum using the photoluminescence yield (PLY) → **element and chemical specificity**

XEOL - Conversion of X-ray energy into optical emission

X-ray phosphor

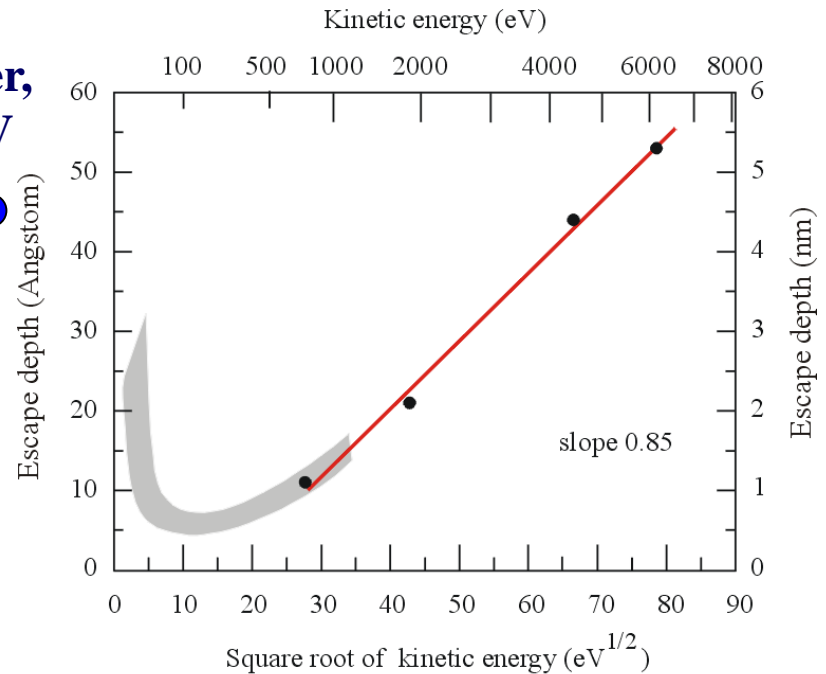
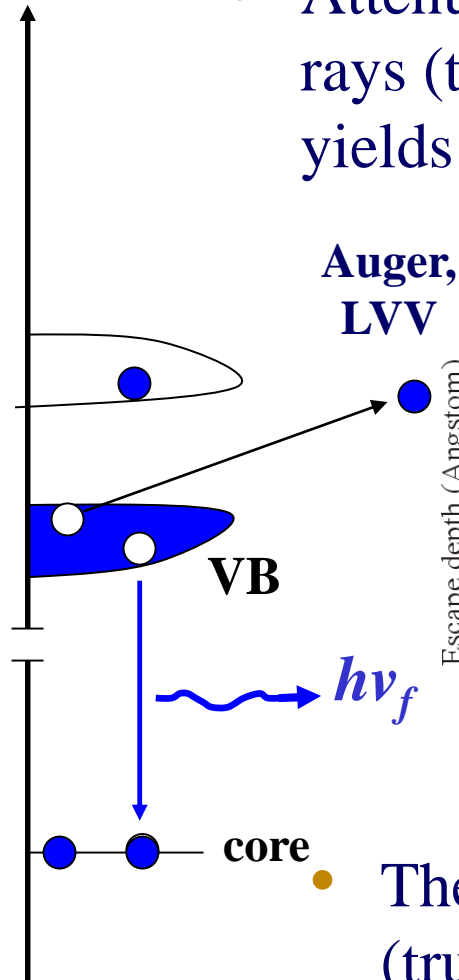
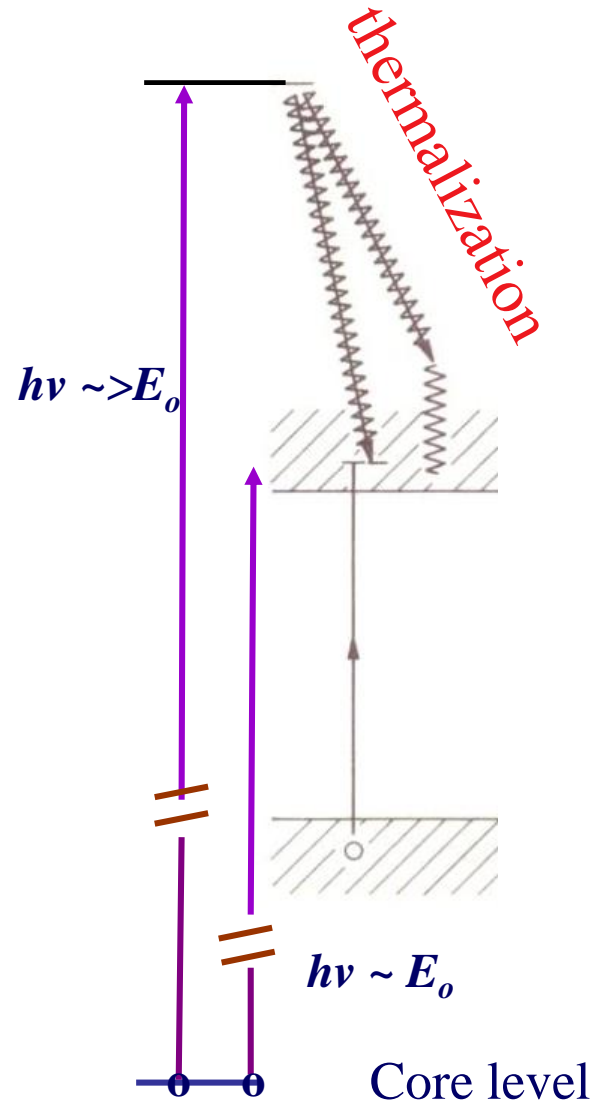


XAS and XEOL (Optical XAFS)



De-excitation, Energy transfer in nanostructures

- Decay channels
- Attenuation of e and fluorescence x-rays (thermalization) in the solid yields secondary electrons (holes)



Thermalization track is confined (truncated) in nanostructures

Photon-in: Synchrotron Light

- • Tunability → Element, edge, excitation channel specificity
- High brightness
- Polarization
- • Time structure

XEOL Techniques

Energy Domain

XEOL: Luminescence induced by selected excitation photon energy (usually across an absorption edge)

Optical XAFS: Monitored with the optical signal

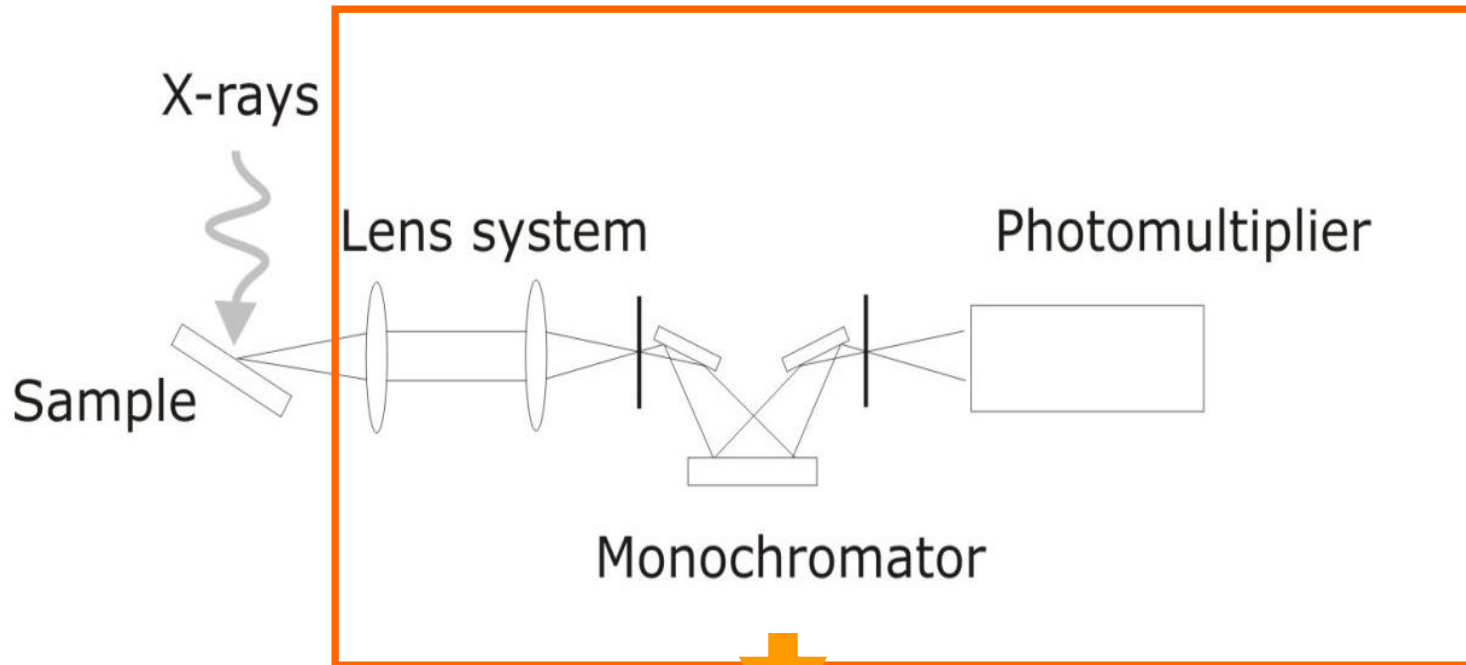
Time Domain

Lifetime: Synchrotron pulse

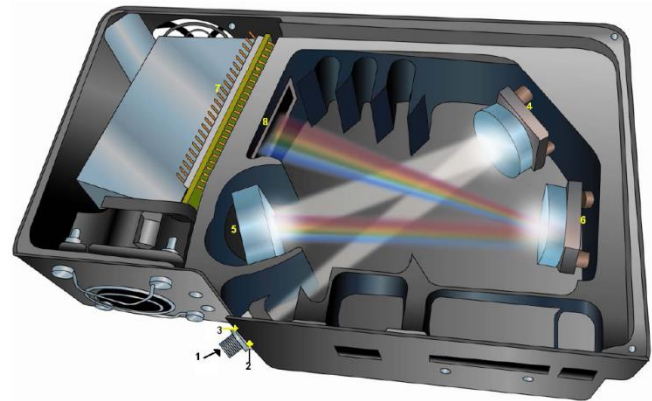
Time-resolved (gated) XEOL: Luminescence within a selected time window between pulses

Time-gated Optical XAFS: Time window

XEOL: Experimental Layout



Alternative: fiber optics,
spectrograph with CCD
detectors (e.g. Ocean
Optics QE65000)



QE65000 Spectrometer with Components

XEOL - Energy Domain

- Case studies

- Si nanostructure

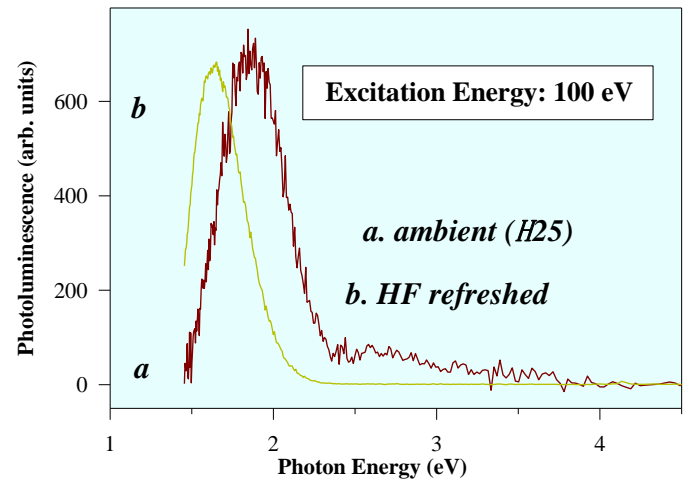
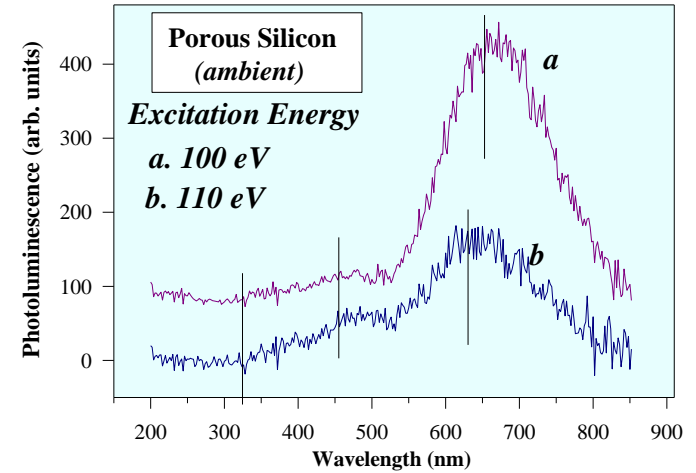
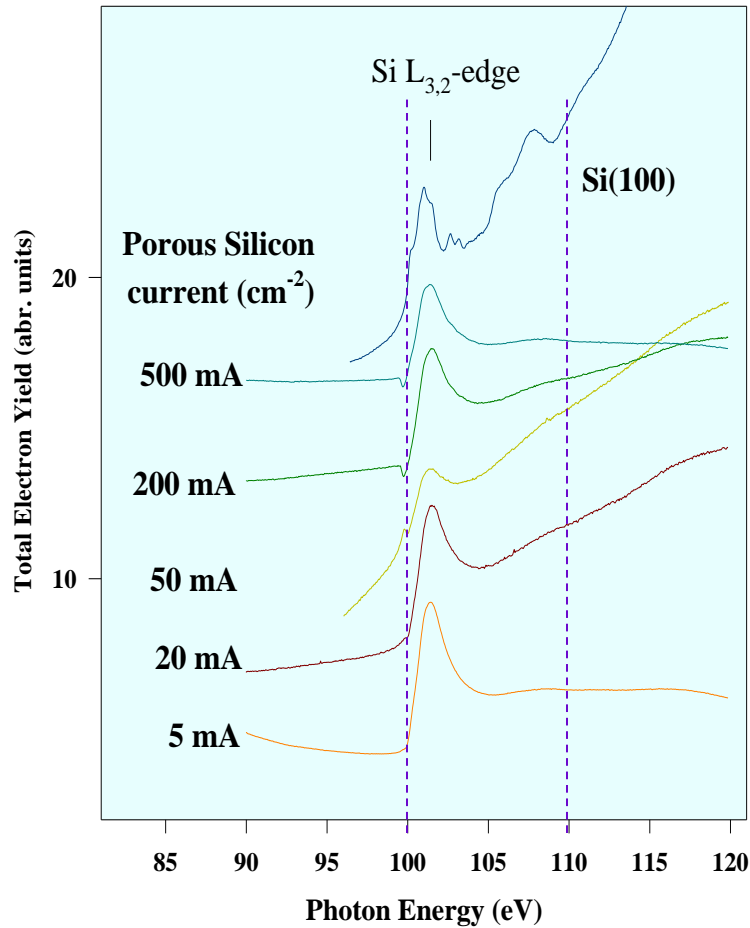
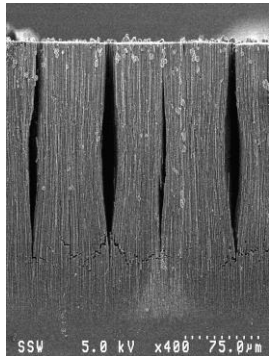
- ZnS nanoribbons (crystal structure engineering)

- Soft matter

- Alq₃ (OLED materials)

- 2D – XEOL TiO₂ nanowire

Si L-edge XEOL (porous silicon)



Si K-edge XEOL of silicon nanowires

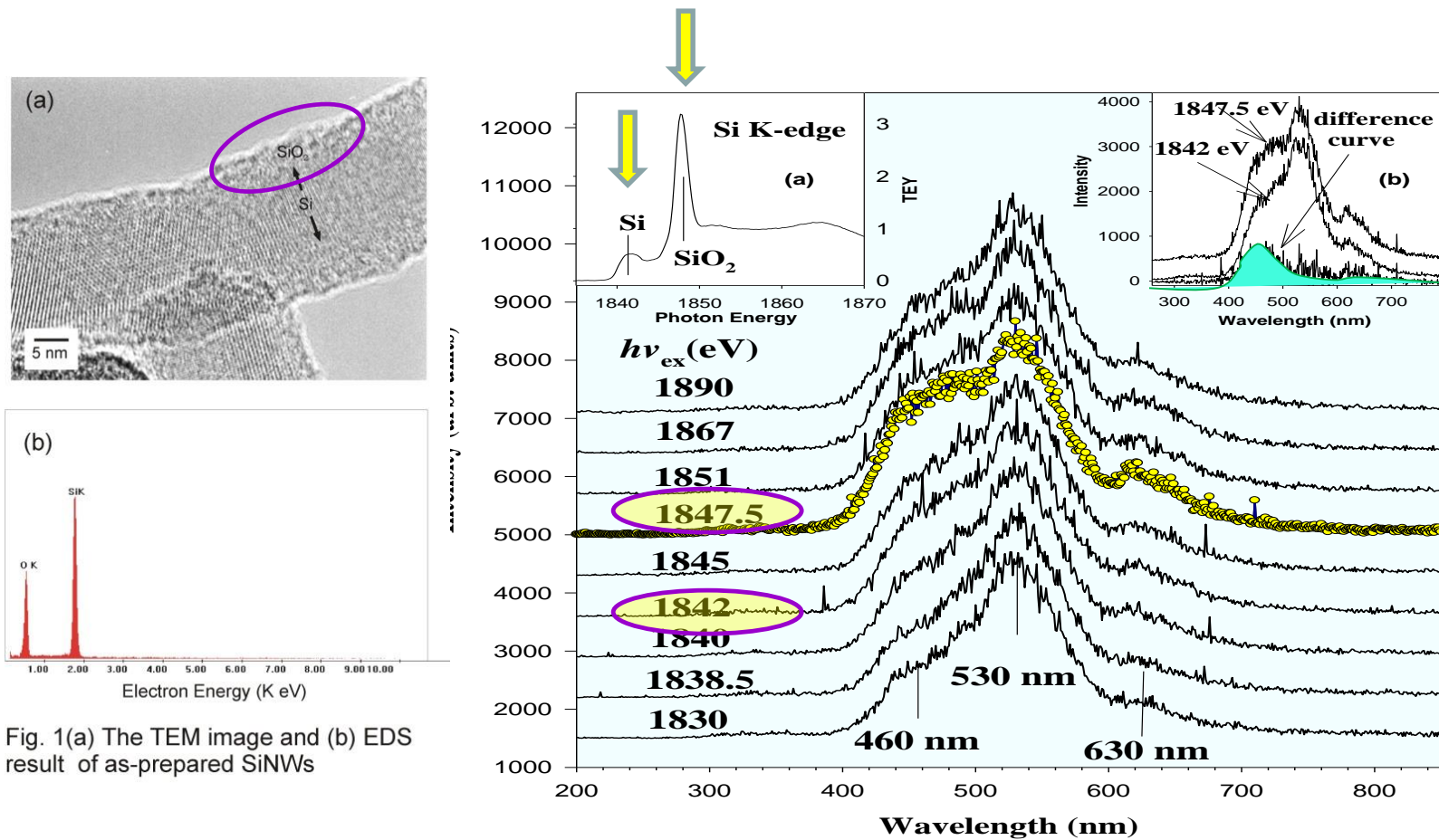
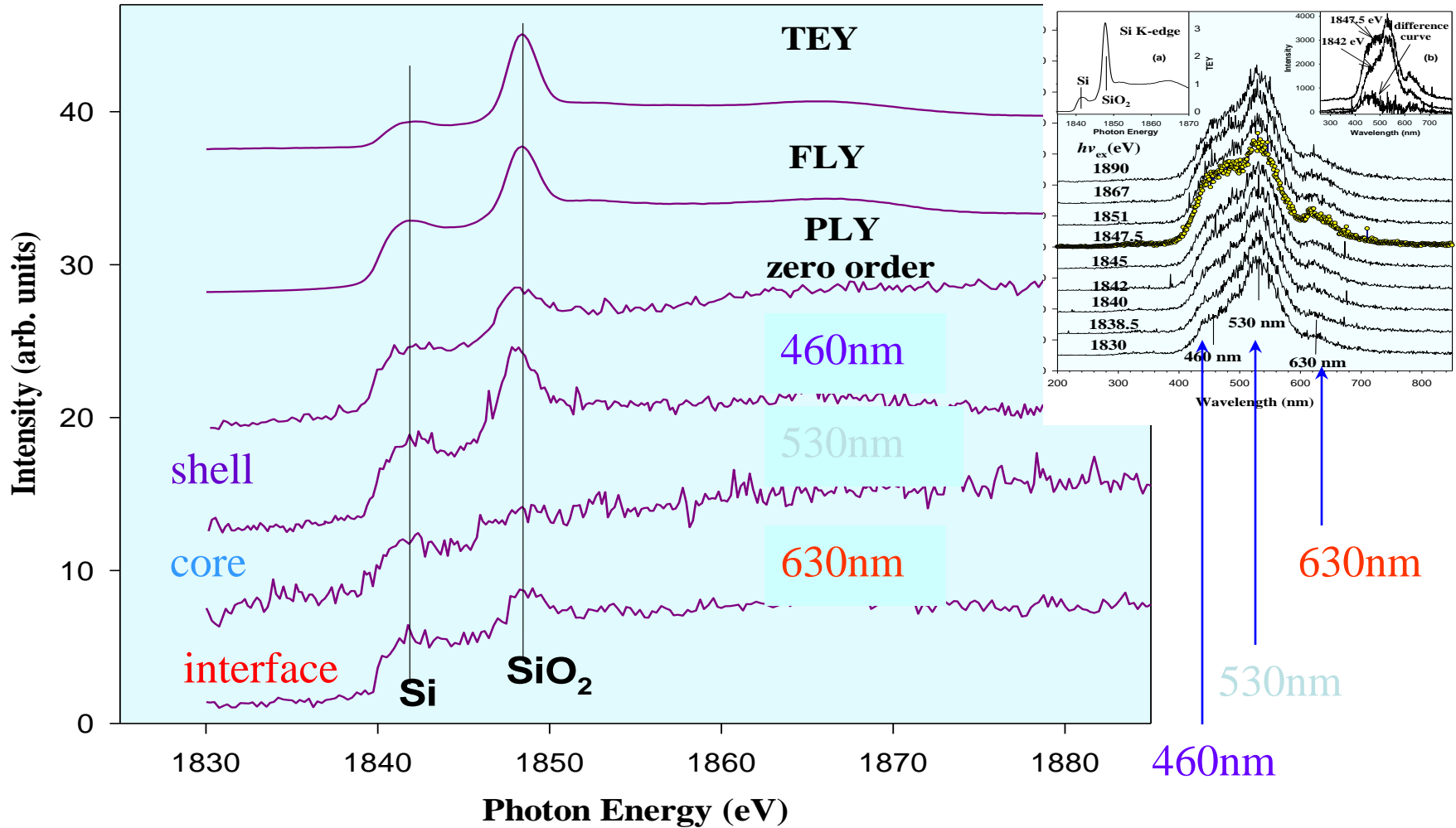


Fig. 1(a) The TEM image and (b) EDS result of as-prepared SiNWs

Phys. Rev. B 70, 045313 (2004)

Si K-edge: PLY

Si nanowire



XEOL and chemistry of SiNW

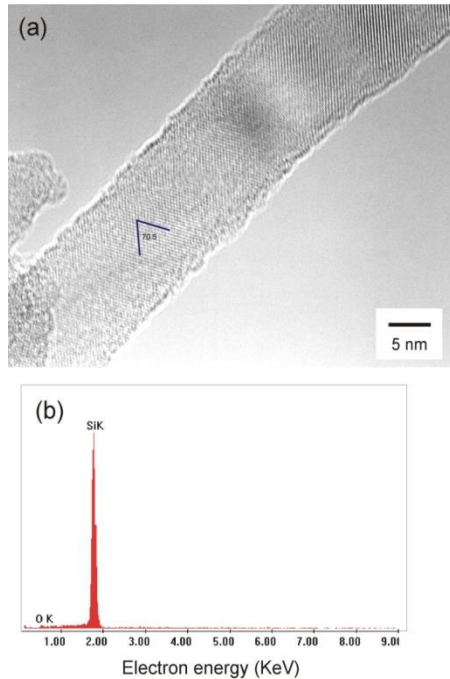
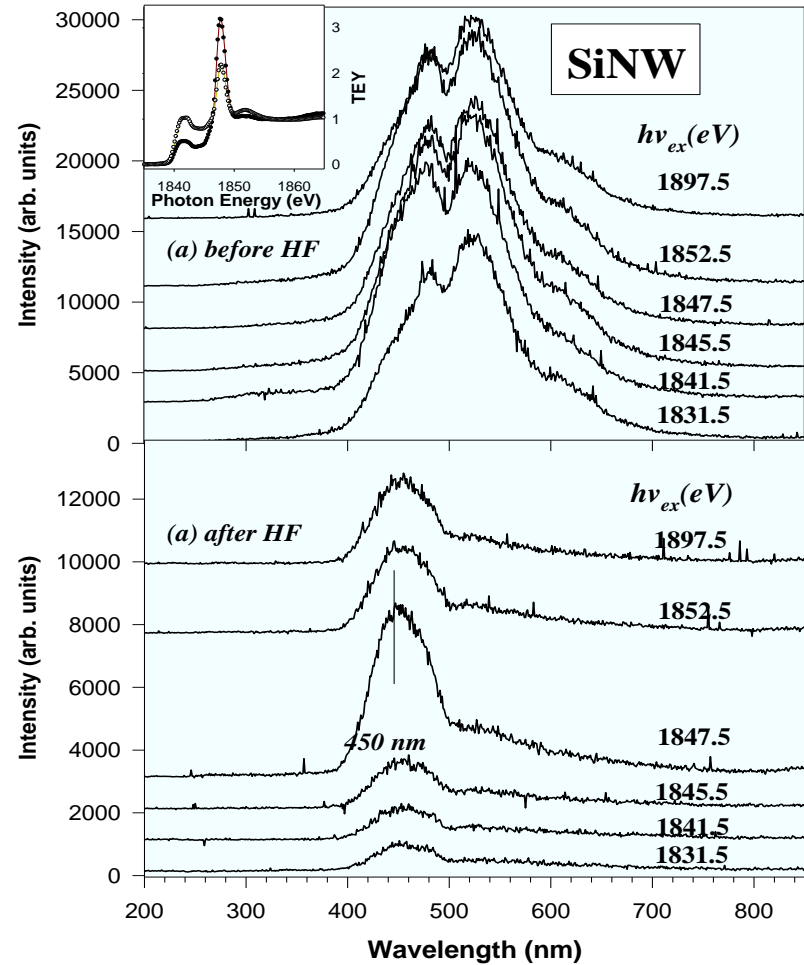
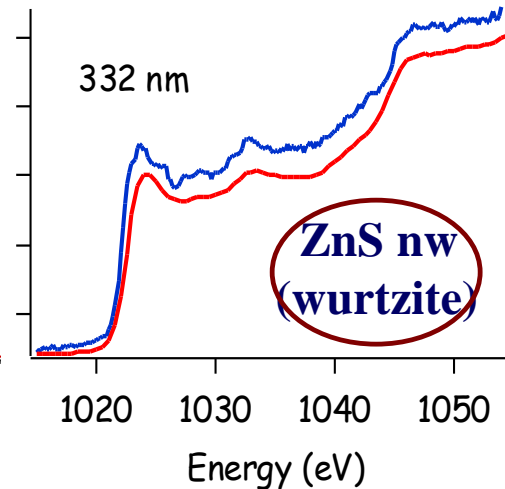
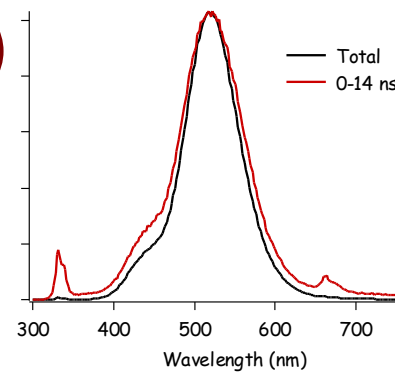
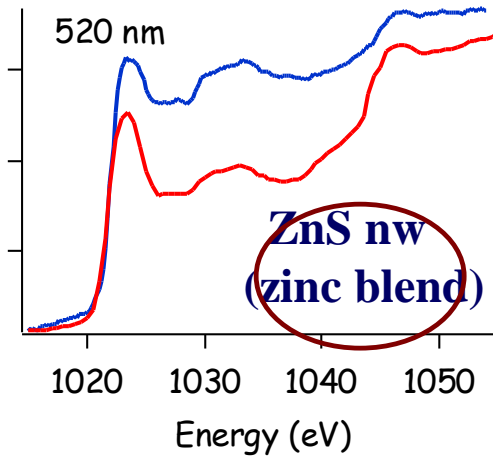
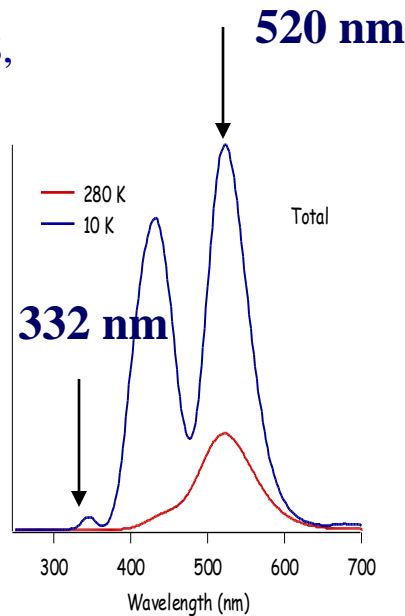
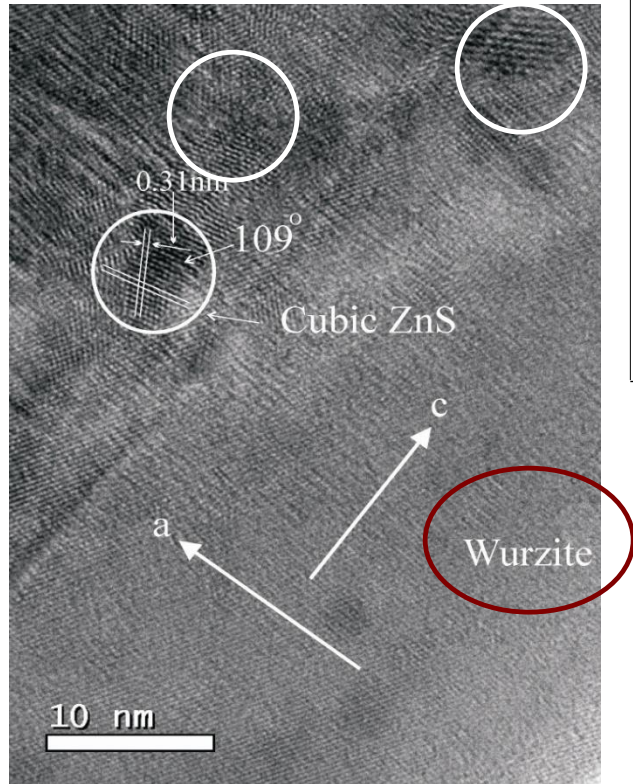


Fig. 2 The TEM image (a) and EDS result (b) of HF etched SiNWs



ZnS hetero-crystalline nano-ribbon

X.-T. Zhou et al., *J. Appl. Phys.* 98, 024312(2005)

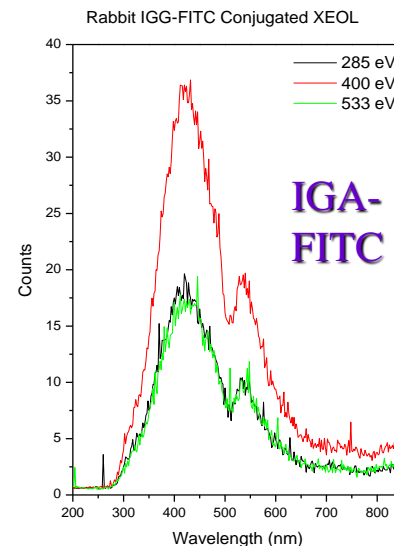
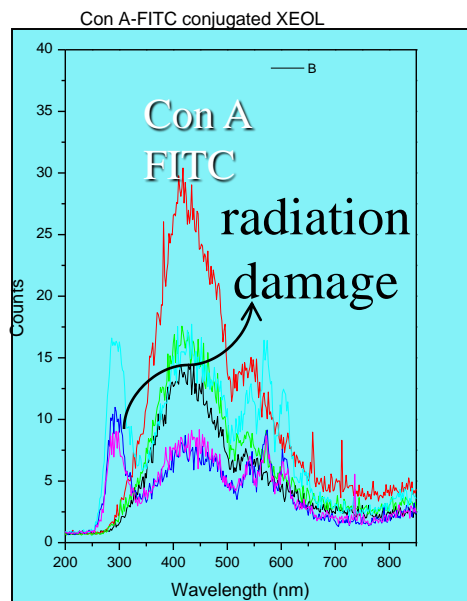
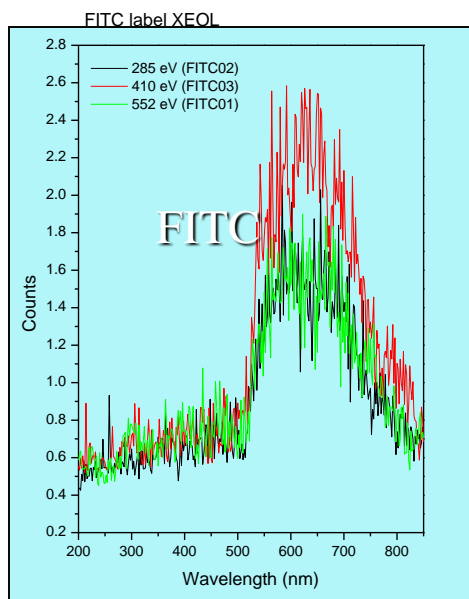
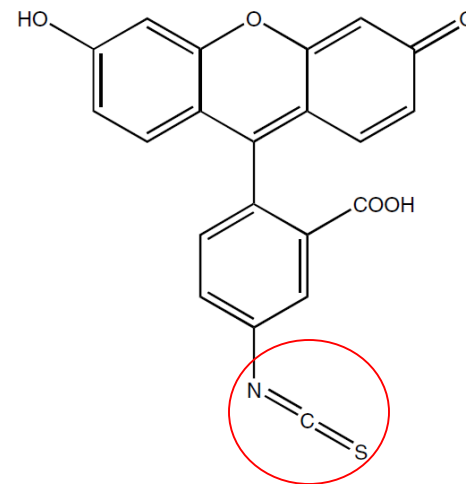


R.A. Rosenberg et al., *Appl. Phys. Lett.* 87, 253105(2005)

XEOL from soft matters

Fluorescein isothiocyanate (**FITC**) label

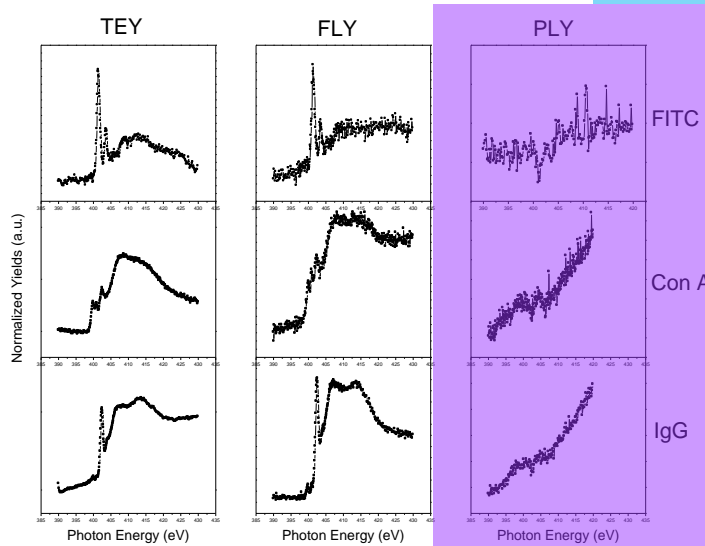
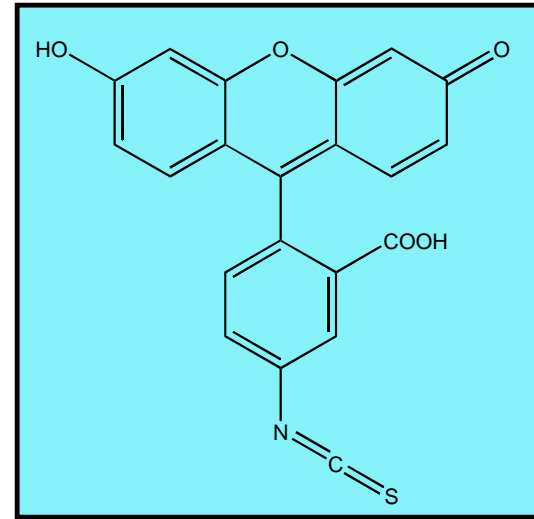
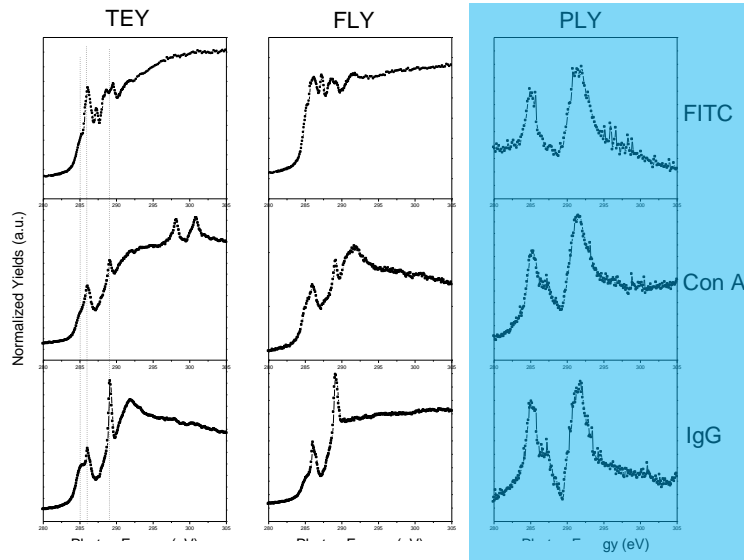
FITC conjugated concanavalin A (Con A) lectin (**Con A-FITC**) and goat anti-rabbit immunoglobulin G (IgG) (**IgG-FITC**)



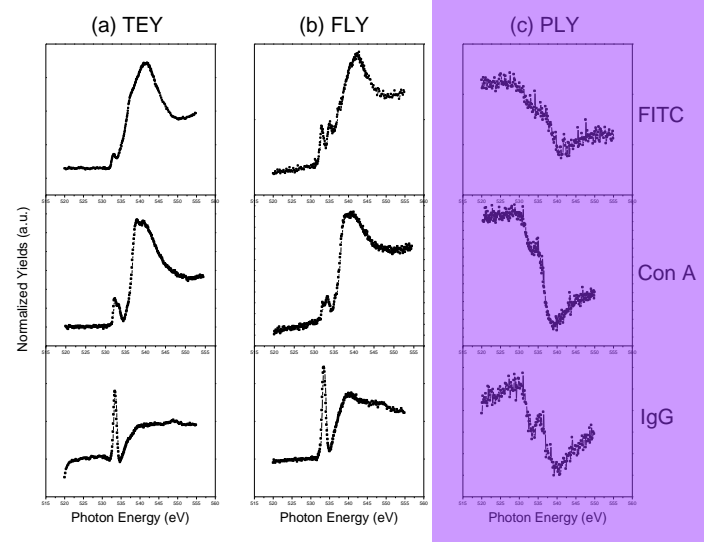
P.-S. G. Kim et al. Chem. Phys. Lett. 39, 44(2004)

PLY from labeled proteins

C K-edge

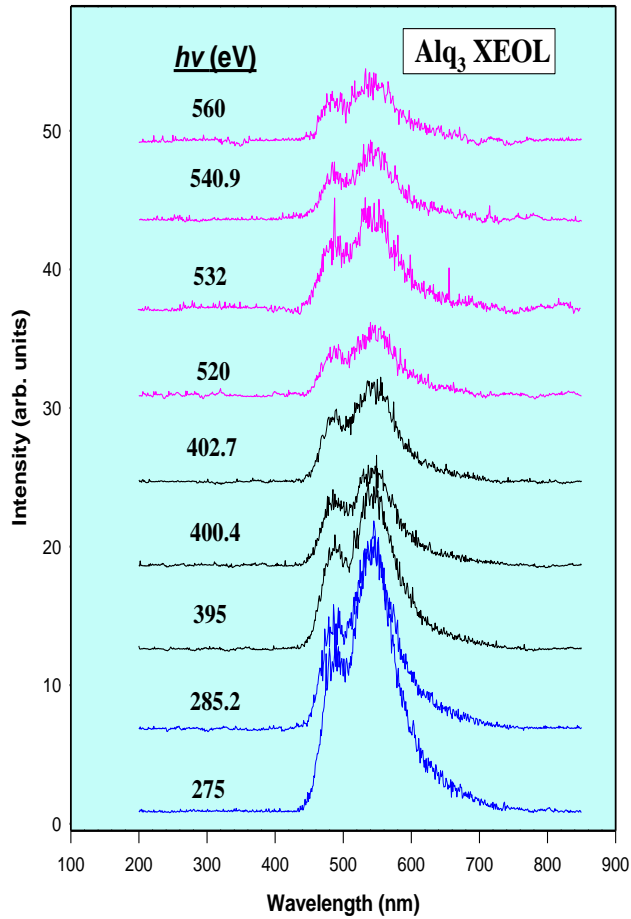


N K-edge

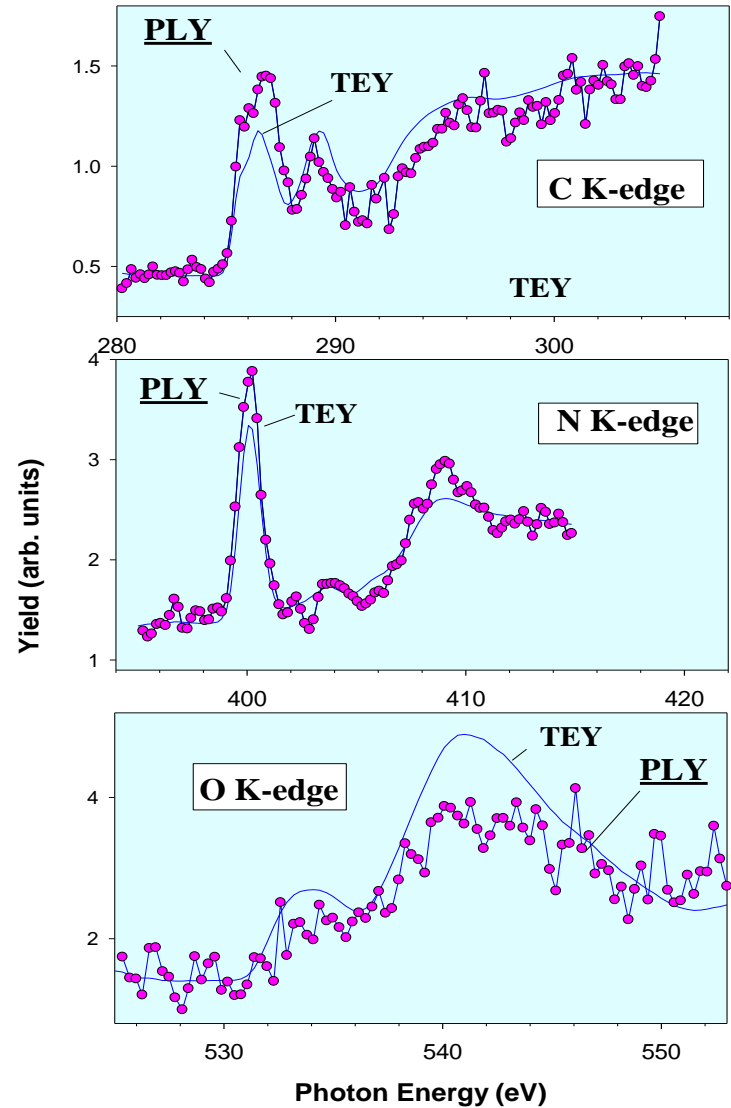


O K-edge

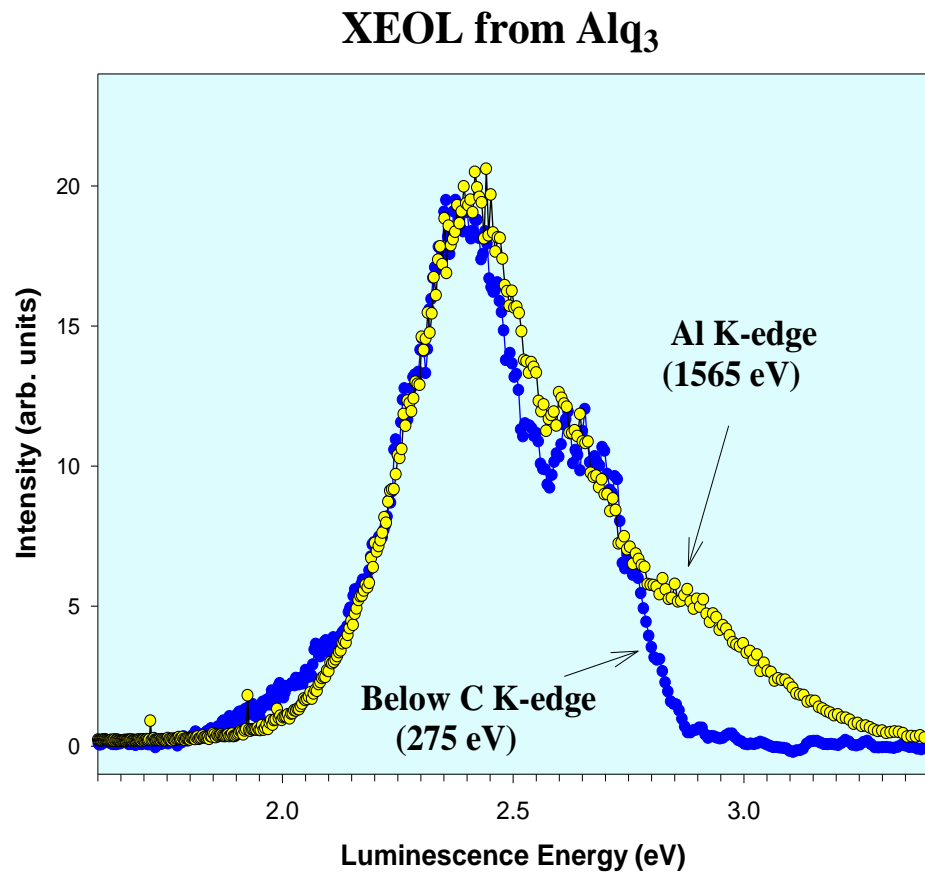
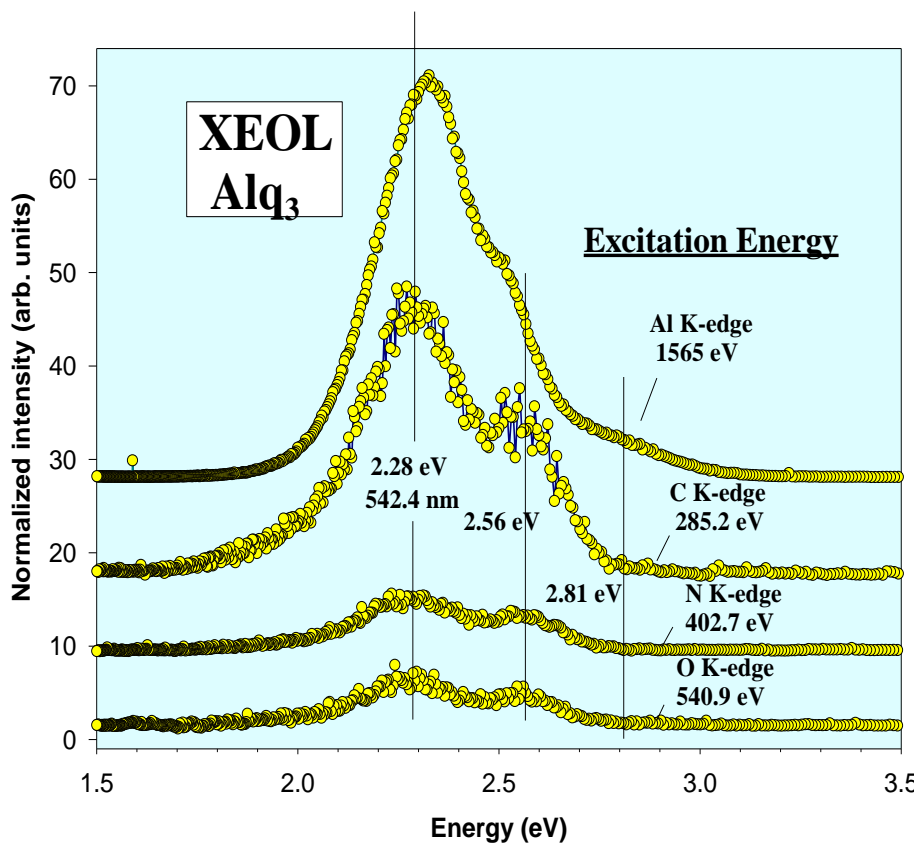
Alq₃ XEOL



XAFS (near-edge)

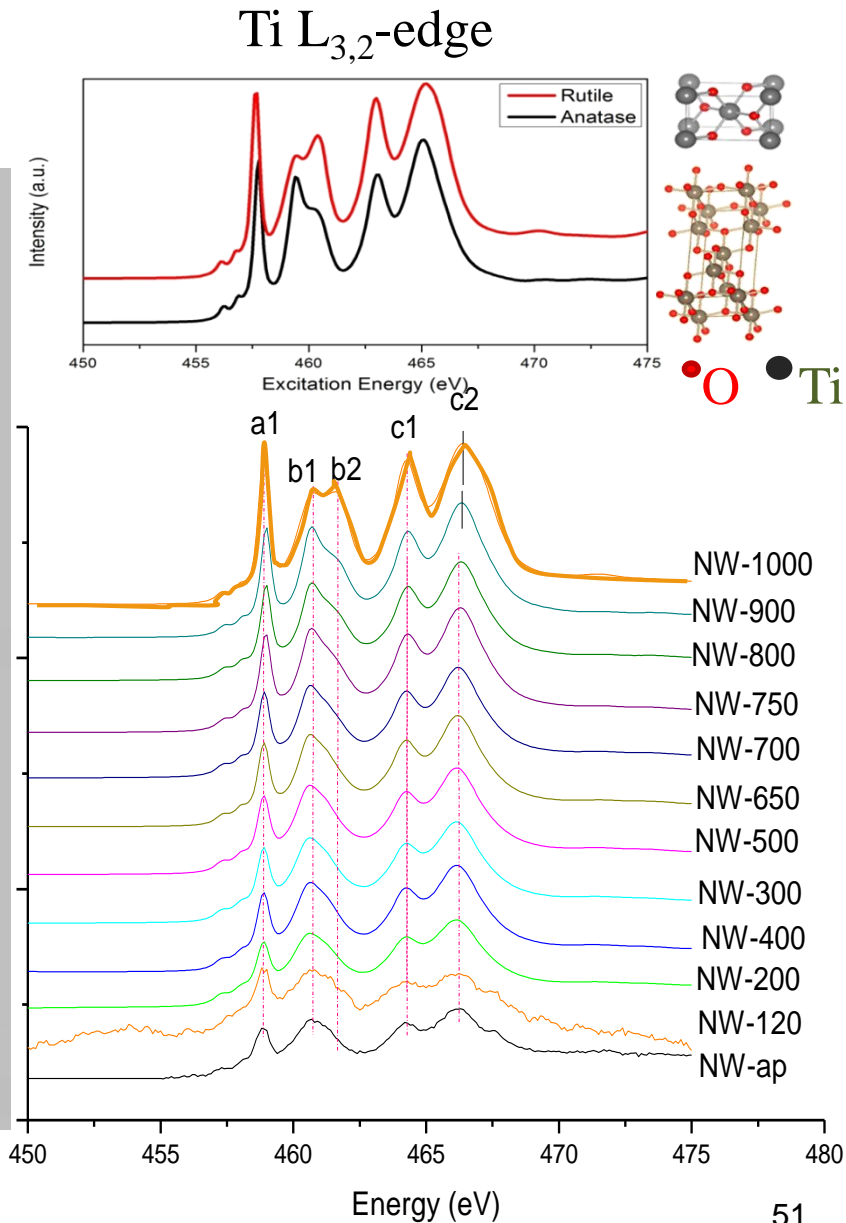
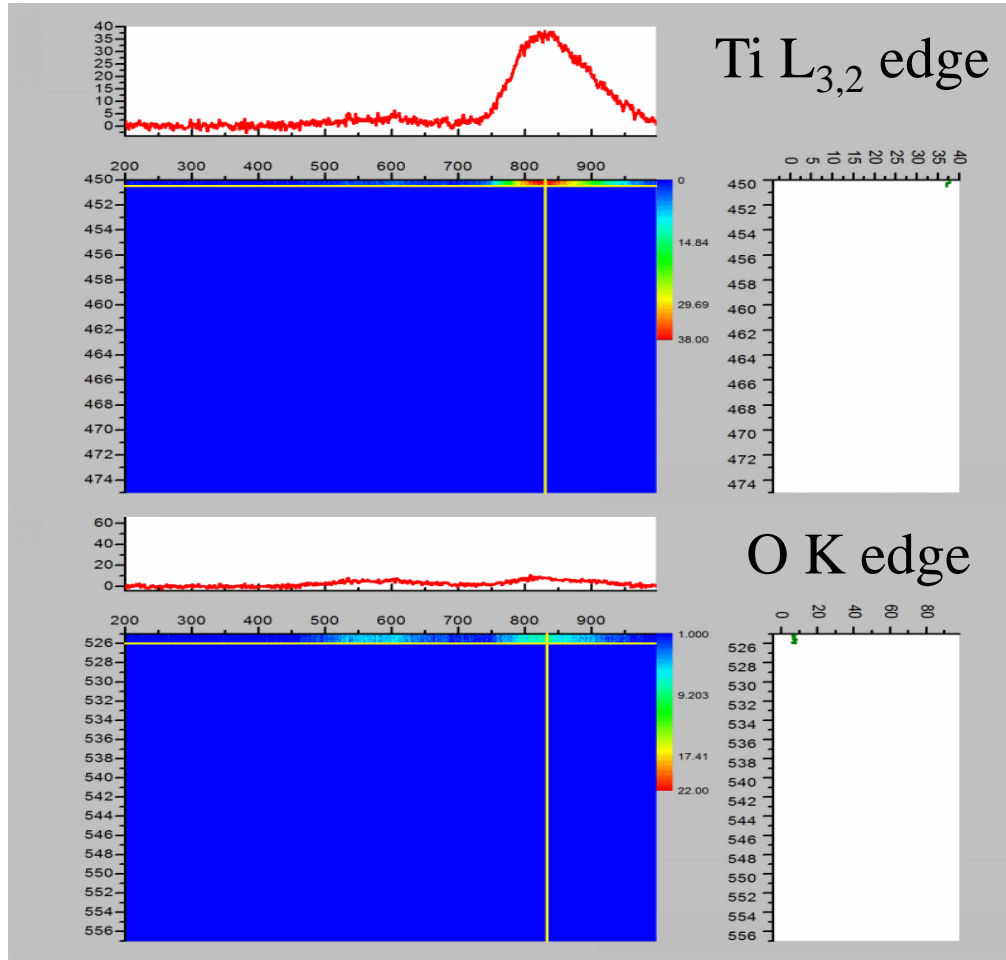


Excitation Channel Dependent XEOL



2-D XAS-XEOL TiO₂ Nanowire (1000 °C)

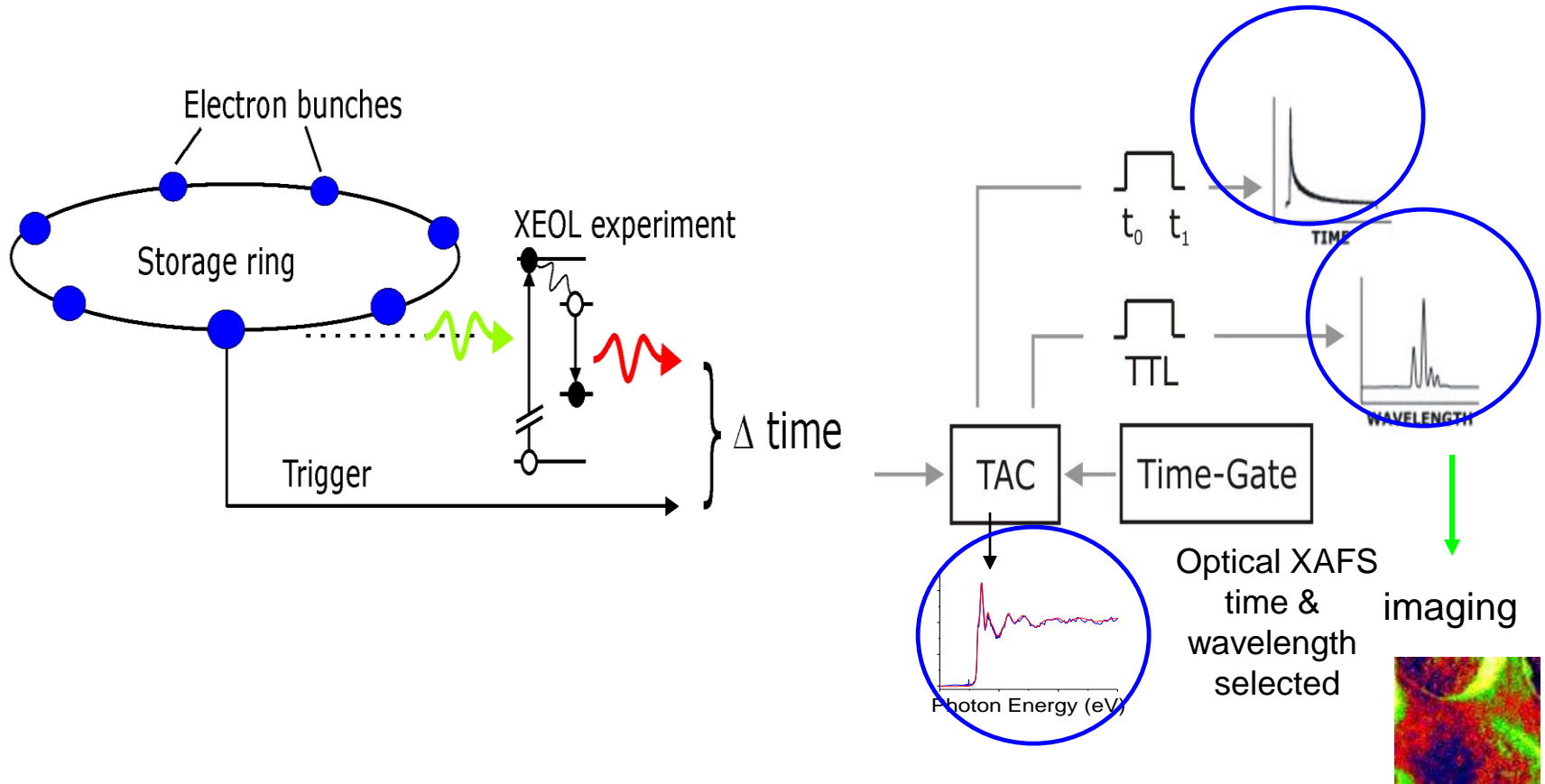
anatase rutile



A. Zhao et al. unpublished

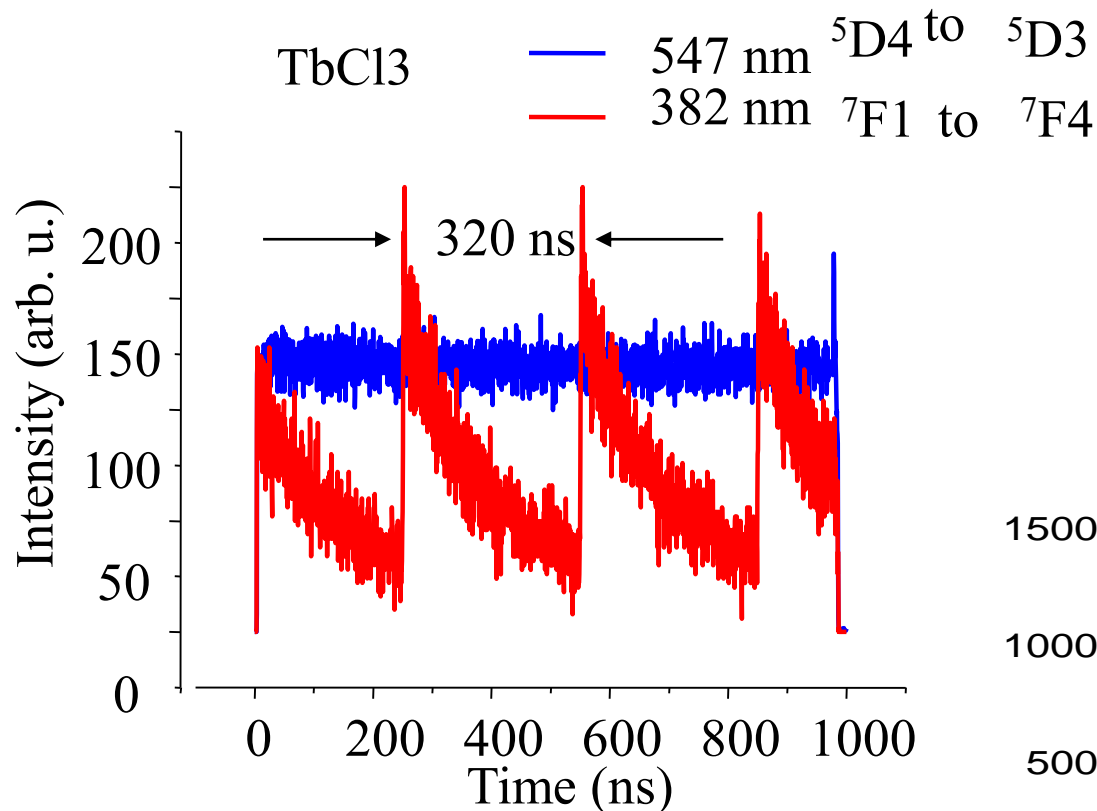
XEOL - Time Domain

TRXEOL-Time-resolved XEOL

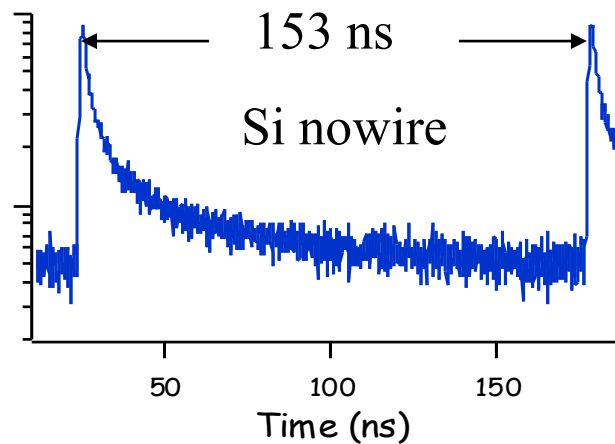


TRXEOL from SRC, APS and CLS

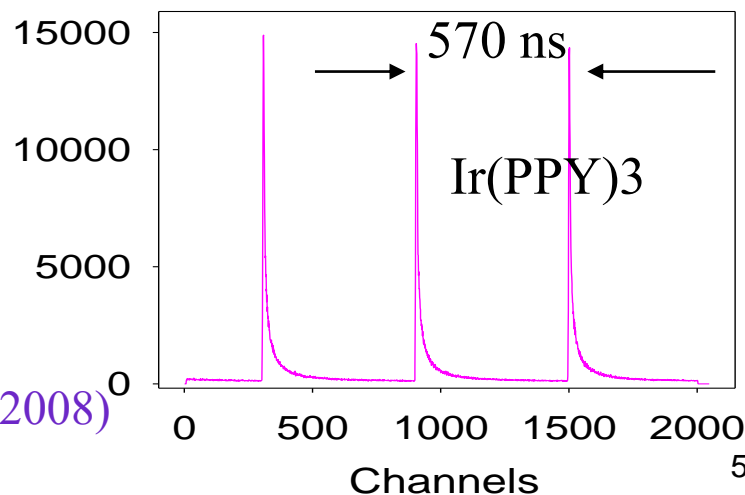
SRC single bunch



APS top up: 24 bunches



CLS single bunch



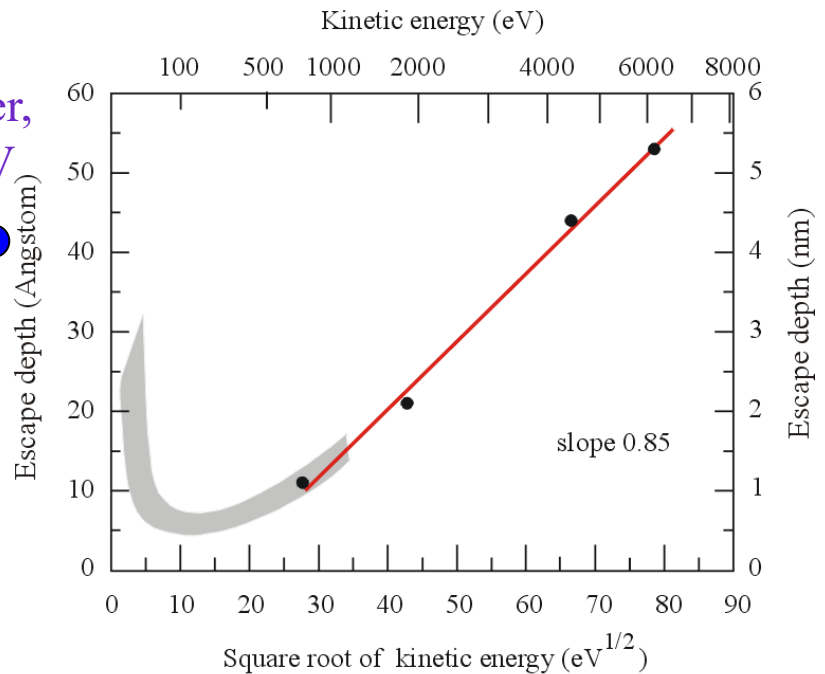
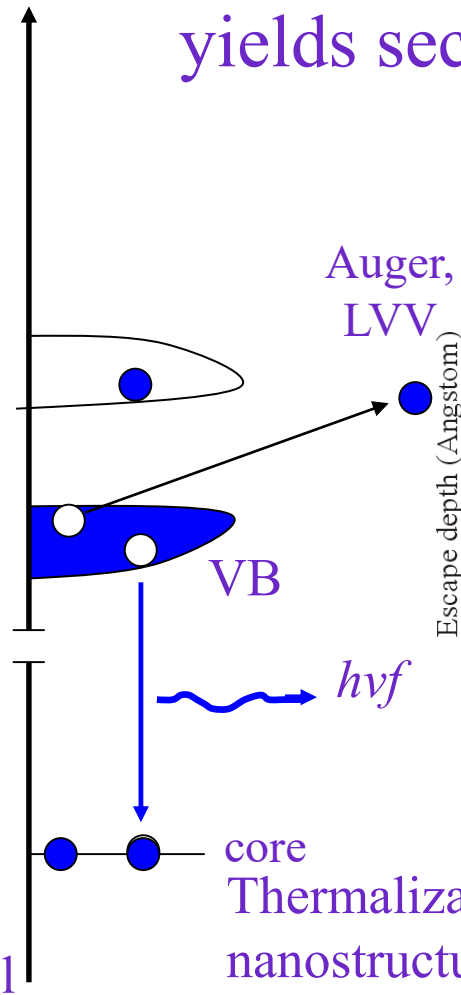
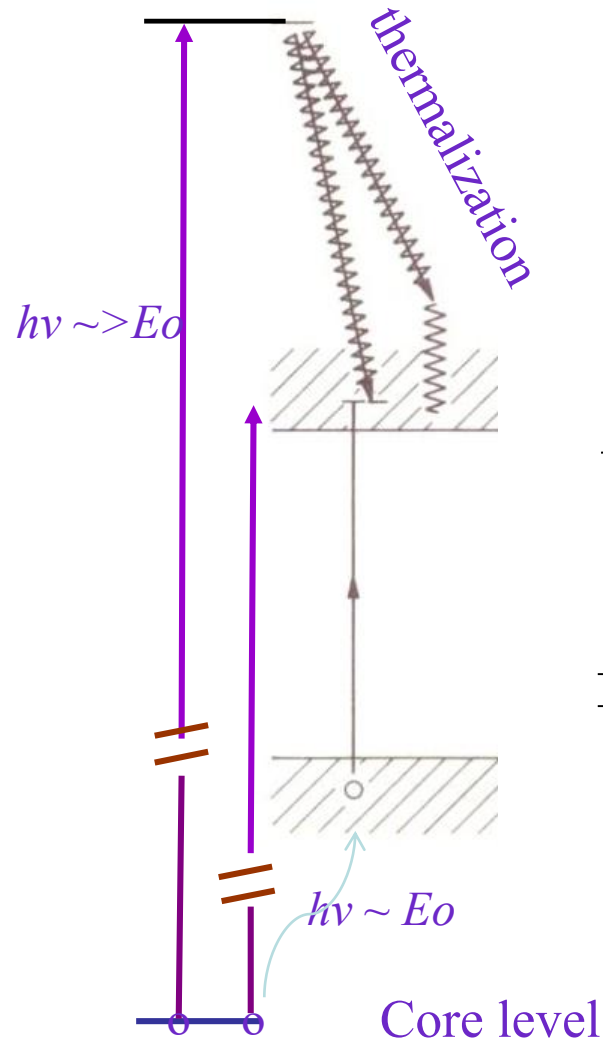
F. Heigl, et.al. SRI2006, AIP CP879, 1202 (2007)

R.A. Rosenberg et. al., J. Phys. Chem., 112, 13943 (2008)

F. Heigl, et al. JACS, 128, 3906 (2006)

De-excitation and energy transfer in nanostructures

- Attenuation of e (thermalization) yields secondary electrons (holes)

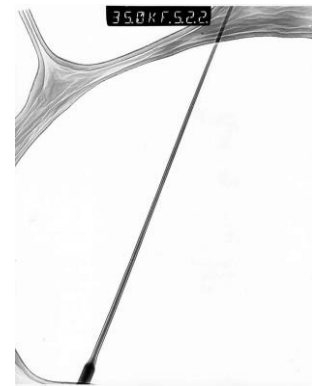


Thermalization track is confined (truncated) in nanostructures ! The smaller the size, the shorter the track, the faster the decay

TRXEOL – Case studies

- ZnO: nanoneedle vs nanowire
(crystallinity)
- ZnO: nano vs micro **(size)**
- Ru(phen)_3^{2+} : **(metal vs ligand)**

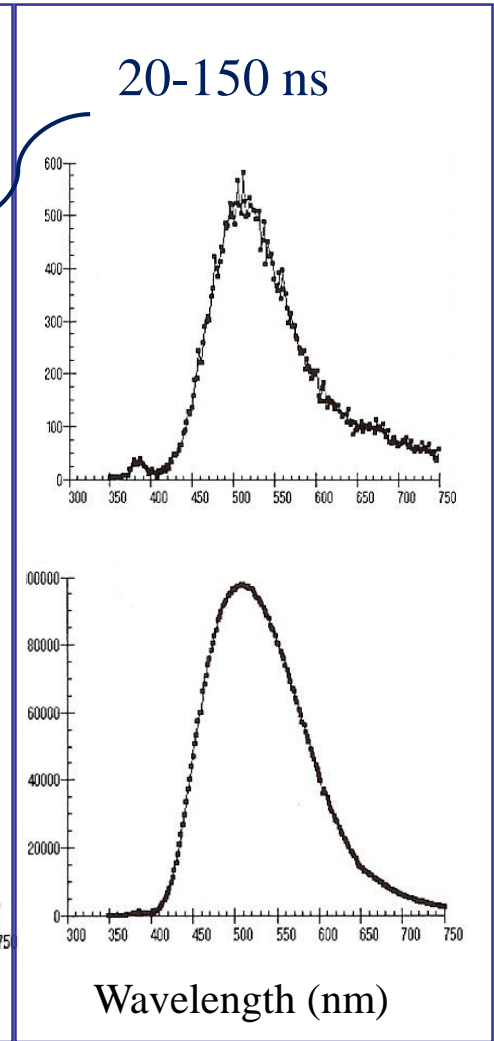
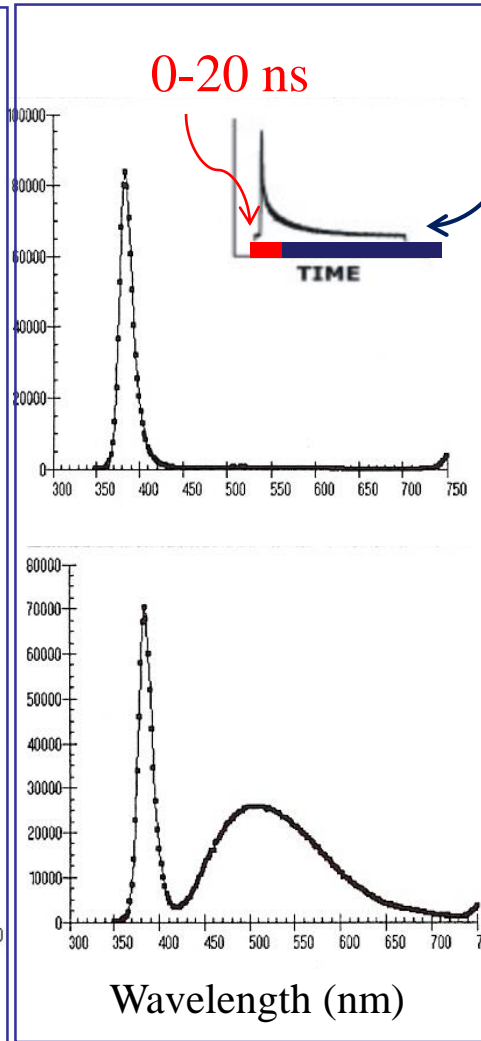
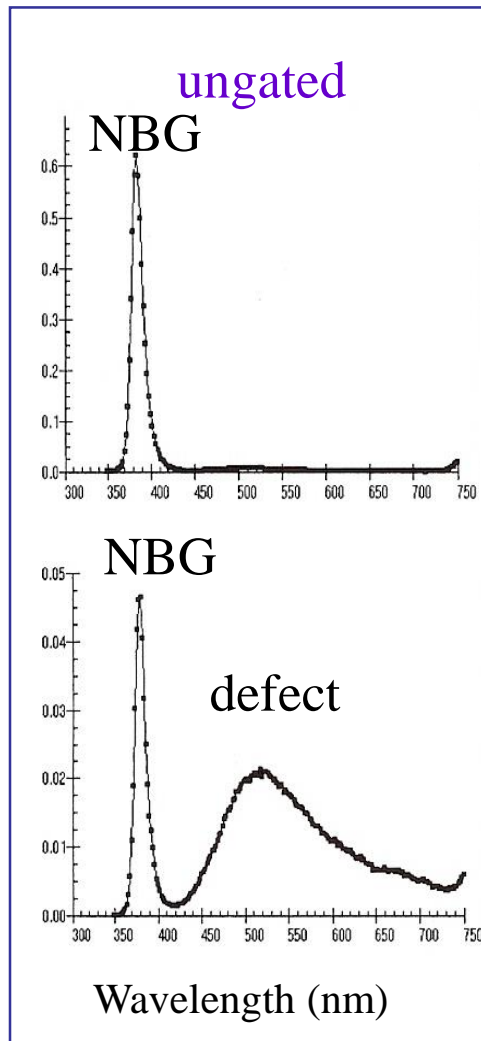
TRXEOL from ZnO nanostructure



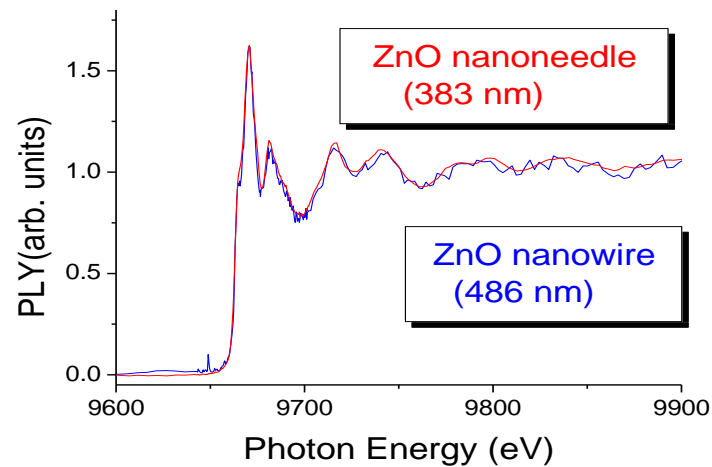
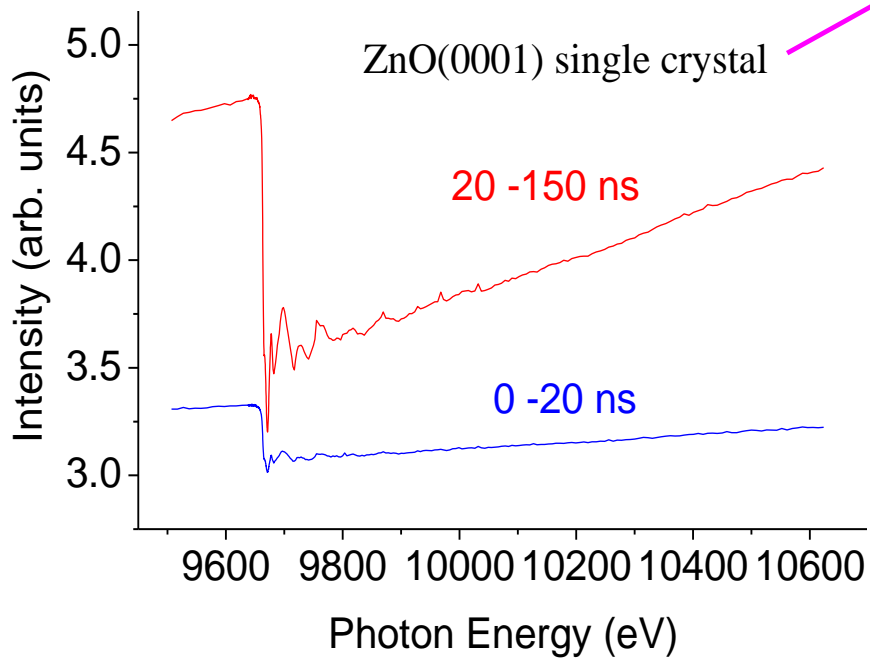
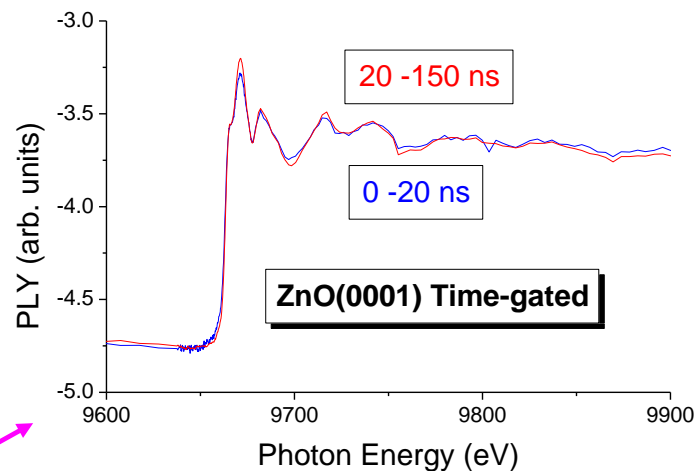
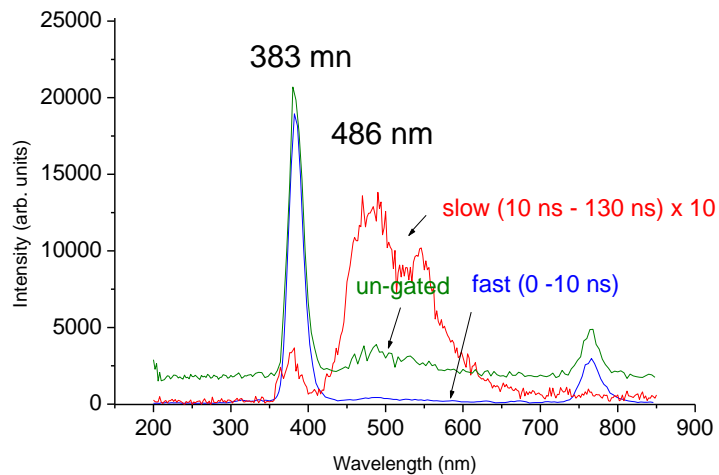
nanoneedle



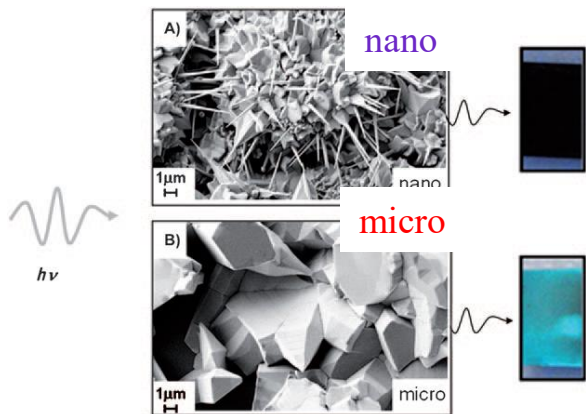
nanowire



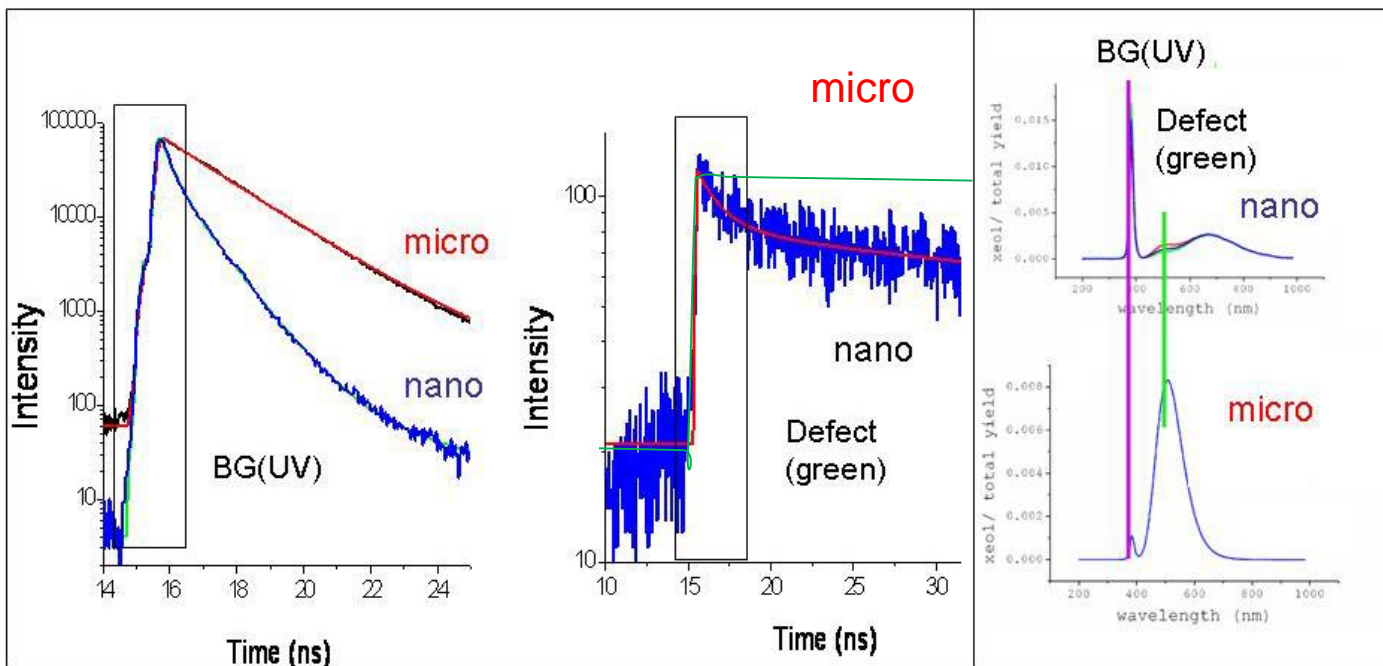
Time gated-optical XAFS



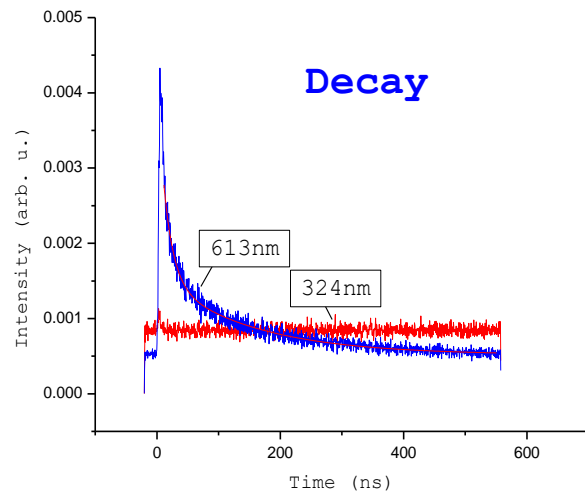
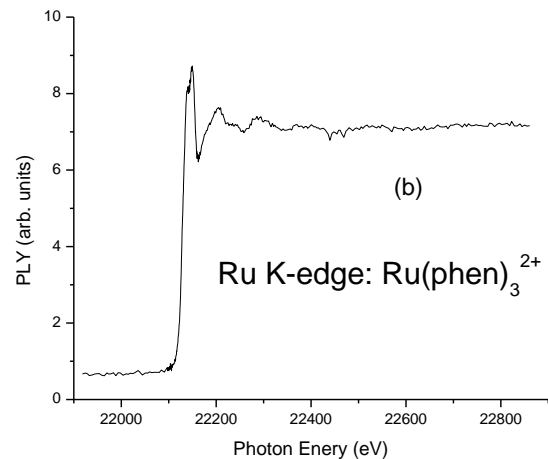
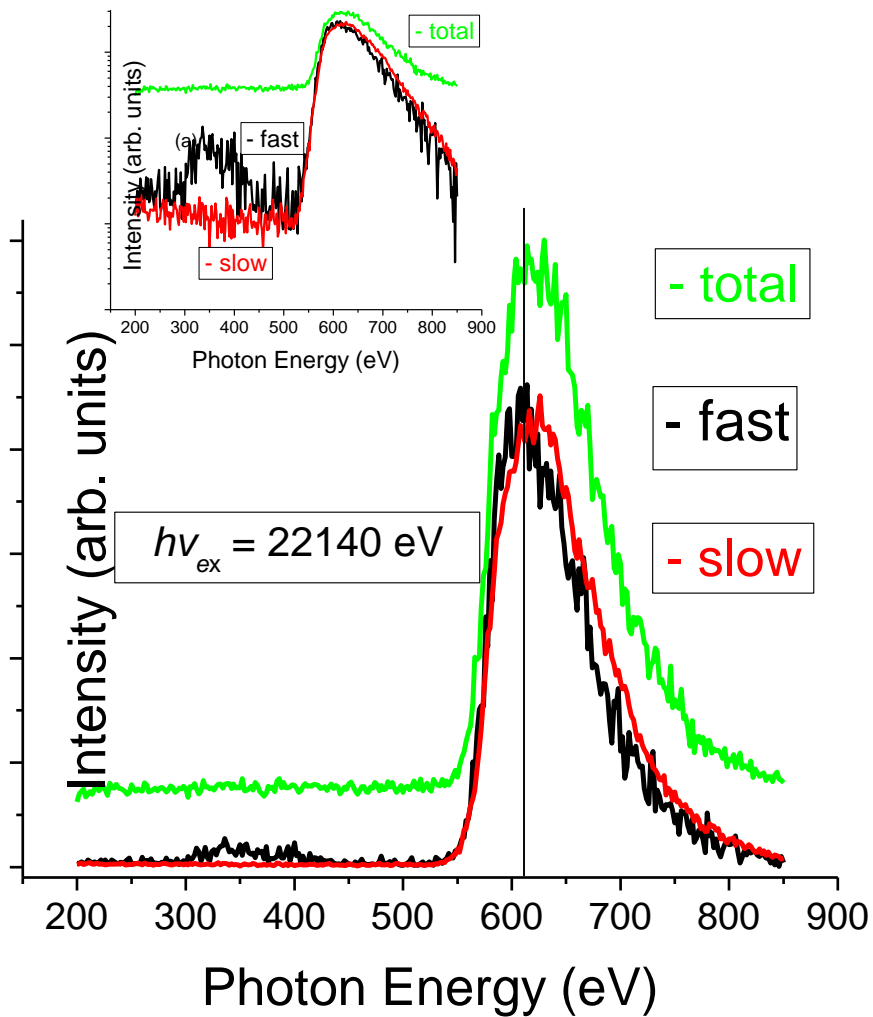
Morphology and size dependent dynamics



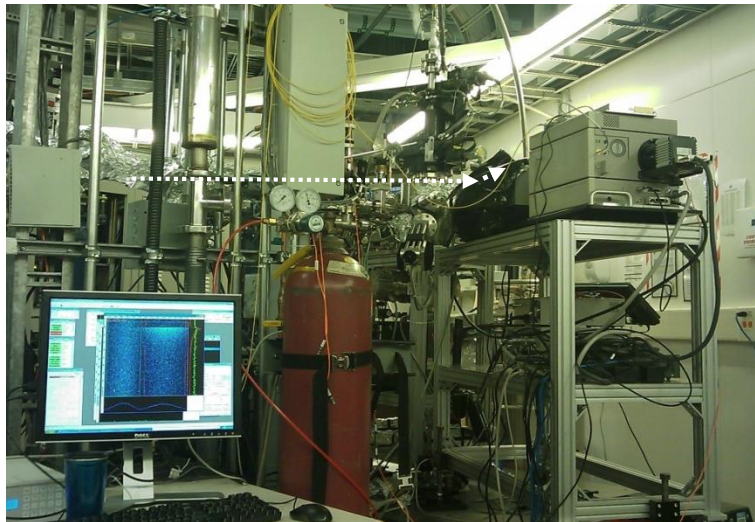
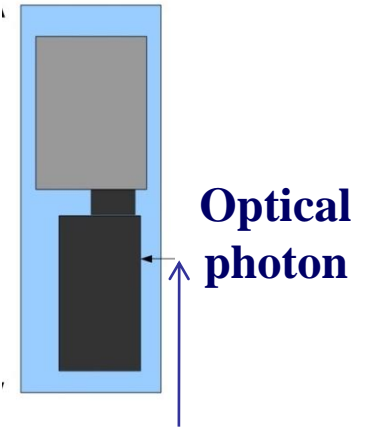
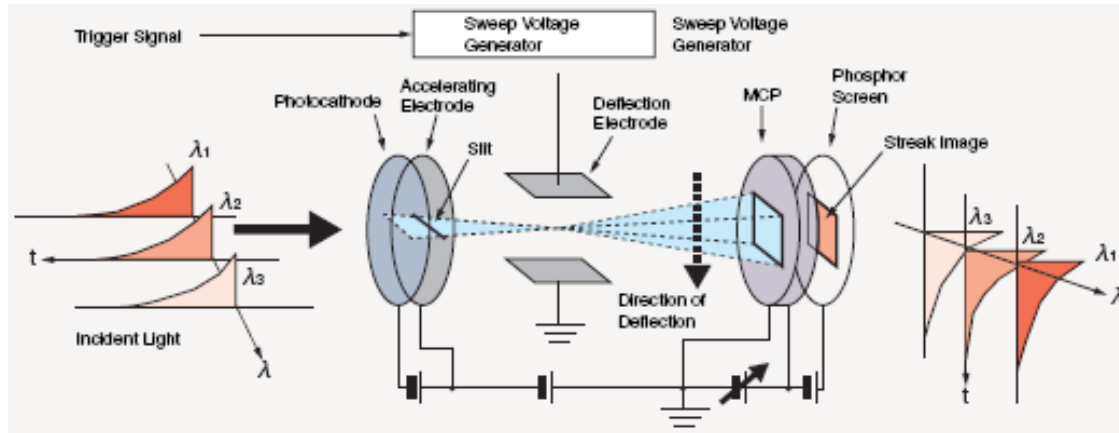
L. Armelao et. al. *ChemPhysChem*
2010, 11, 3625



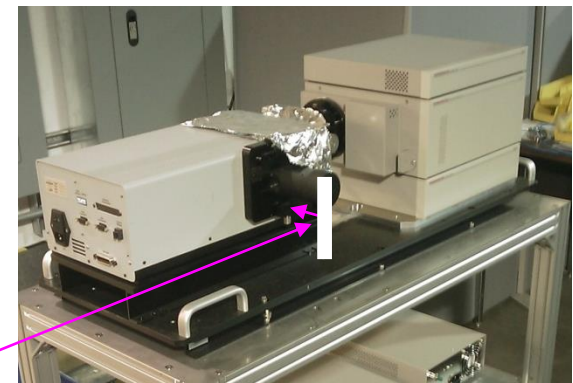
TRXEOL from $\text{Ru}(\text{phen})_3^{2+}$



Prospect: TRXEOL with an optical streak camera



T. Regier et al. *CLS Activity Report*, 2009

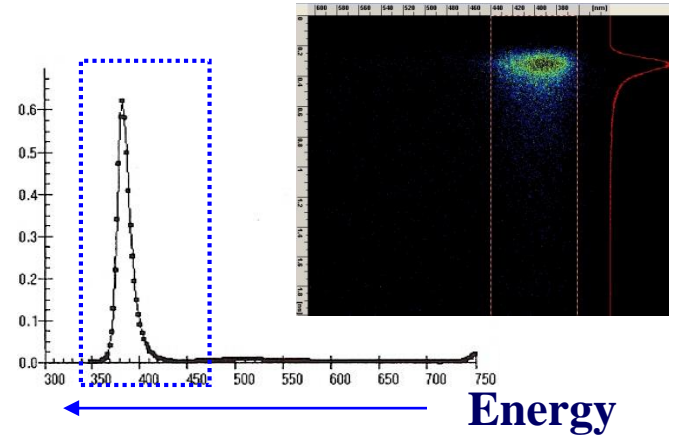
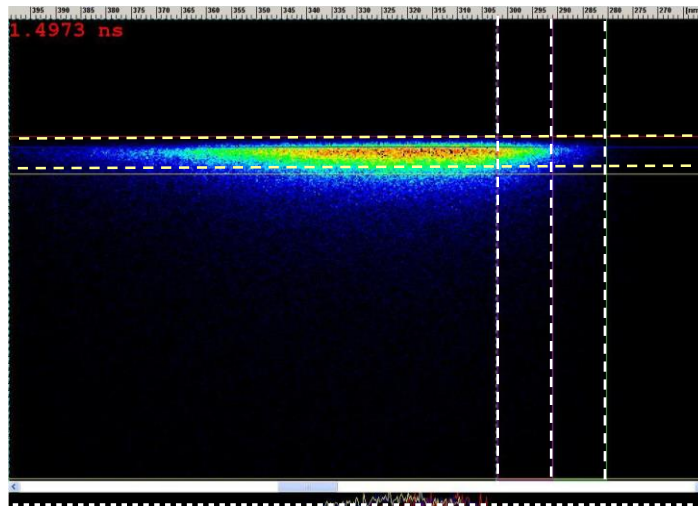


Hamamatsu C4780

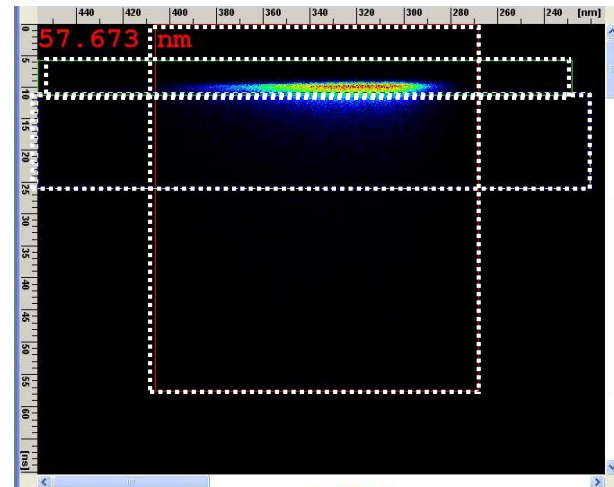
ZnO nanoneedle excitonic emission

$$\lambda_{\max} \sim 380 \text{ nm } (h\nu_{\text{ex}} = 530 \text{ eV})$$

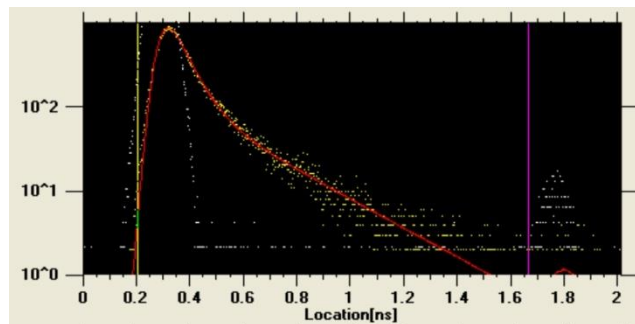
Time

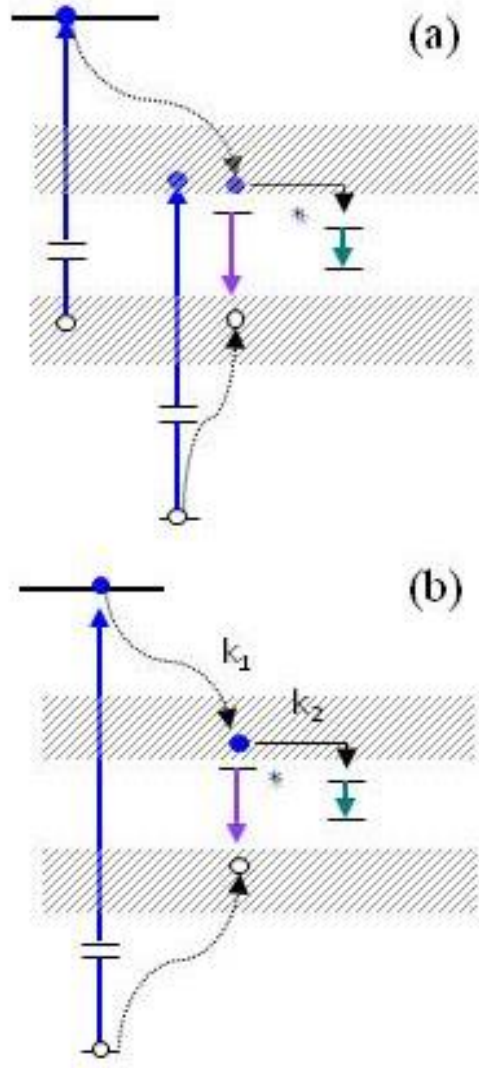


Energy



Energy





What determines the decay lifetime of XEOL ?

In general,

- The larger the energy separation the faster the decay
- The faster the thermalization, k_1 , the faster the decay (nanostructures)
- The faster the energy transfer process, k_2 , the faster the decay
- NBG emission is always faster than defect emission